

The quenching of the Yb (III) porphyrazine complexes and heterocyclic ligands by C₆₀ in solutions

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The interaction of the Yb(III)tetra-aryl-tetra-cyano-porphyrazine complexes (Aryl=C₆H₅(I), C₆H₄F(II), C₆F₅(III), C₁₀H₇(IV)) with C₆₀ had been investigated. Complex (I) is readily soluble in organics. It is well compatible with a variety of polymeric matrices giving doped polymeric glasses, films and solutions which are highly luminescent ($\phi = 13\%$) in the biologically relevant optical window covering the visible and near infrared range.

We investigated complexes (I-IV) luminescent spectra ($\lambda_{\text{excitation}} = 300 \text{ nm}$) in acetonitrile solutions using C₆₀ as a quencher in visible region ($\lambda_{\text{emission}} = 420\text{-}460 \text{ nm}$). It had been found [1] that a quenching was caused by resonance interaction of aromatic hydrocarbons with fullerene C₆₀. Relative intensity of complexes (I-IV) emission dependence on quencher molar concentration is satisfied to Stern-Volmer equation with $K_q = (1,24\text{-}1,89) \cdot 10^5 \text{ l/mol}$ because of a weak CT-complex formation. The results are shown in the table.

| Aryl | C ₁₀ H ₇ | C ₆ H ₅ | C ₆ H ₄ F | C ₆ F ₅ |
|--|--------------------------------|-------------------------------|---------------------------------|-------------------------------|
| $\lambda_{\text{emission}}(\text{nm})$ | 428 | 456 | 425 | 453 |
| $K_q \cdot 10^{-5}(\text{l/mol})$ | 1.24 | 1.56 | 1.75 | 1.89 |
| R ² | 0.892 | 0.997 | 0.993 | 0.994 |

K_q – Stern-Volmer quenching constants

R² – correlation coefficient.

A comparison of the complex (I-IV) with organic ligands shows one order of magnitude increase of quenching constants. Such the compounds can be potentially used in photovoltaic devices.

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[1] R.G.Bulgakov, D.I.Galimov, *Russ.Chem.Bull.* **3**, 432-436(2007).