

## High-temperature transitions of fullerite C<sub>60</sub> at moderate pressures

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The transitional pressure-temperature phase diagram of fullerite C<sub>60</sub> (who is the metastable phase with respect to diamond and graphite) is quite complicated, since it includes the transition to polymolecular, amorphous and nanocomposite phases [1, 2]. Although the transitions of C<sub>60</sub> under pressure were studied during more than one decade, there are still grey areas on the transitional diagram of C<sub>60</sub>, even at low pressures. By quenching experiments, we have study transitions of fullerite C<sub>60</sub> at high temperatures (1000–1600K) and low/moderate pressures (0.1–1.5 GPa) attainable for large-volume high-pressure apparatus. Particularly we have prepared a new nanocarbon phases, whose structure is drastically differs from those of nanographite phases prepared from C<sub>60</sub> at high temperatures at pressures 1.5-10 GPa. The study of the structure (x-ray diffraction and TEM), EELS spectra, Raman spectra, hardness and elastic moduli of the new phases provides a comprehensive identification of the obtained nanostructured carbon samples. It have been shown that these materials have a high (~90%) elastic recovery, fairly high hardness  $H \approx 10-15$  GPa and record high values of the hardness-to-Young modulus ratio  $H/E \approx 0.17-0.22$ . The latter shows that the observed hardness value is close to the “ideal” limit of the  $H/E$  ratio. The nanostructure of the new carbon materials is thought to represent a combination of interlinked curved fragments of the C<sub>60</sub> molecules and nanographite nuclei.

[1] B. Sundqvist, *Advances in Physics* **48**, 1 (1999).

[2] V.V. Brazhkin and A.G. Lyapin, *New Diamond and Frontier Carbon Technology* **14**, 259 (2004).