



Ioffe Physical-Technical Institute

Department of Theoretical Astrophysics

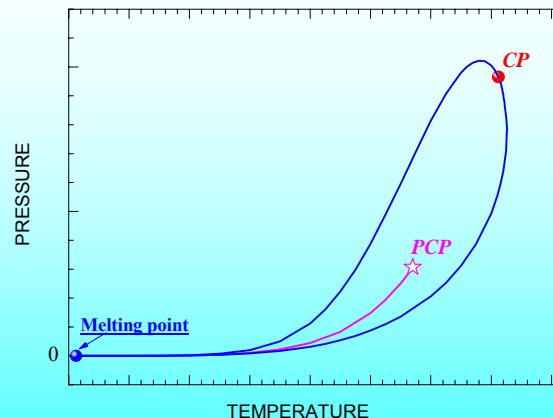
April 17, 2007



NON-CONGRUENT PHASE TRANSITIONS IN INTERIORS OF GIANT PLANETS

Igor Iosilevskiy

Moscow Institute of Physics and Technology
(State University)



1967

Nature 215 (1967);

Internal Structure and Energy Emission of Jupiter

R. Smoluchowski

Princeton University Princeton, New Jersey

Jupiter emits much more energy than it absorbs. Explanations of the source of this heat depend upon our knowledge of its interior and of the behavior of condensed matter at very high temperatures and pressures.

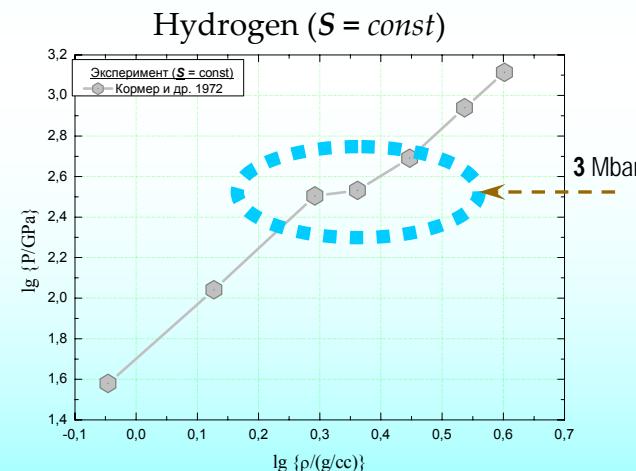
1968 -1970

Norman & Starostin, *Plasma Phase Transition Concept*

1972

Kormer et al. (Russian Nuclear Center (Sarov)),
Density jump in quasi-isentropic compression of hydrogen
($P \sim 3$ Mbar)

? - Plasma Phase Transition - ?

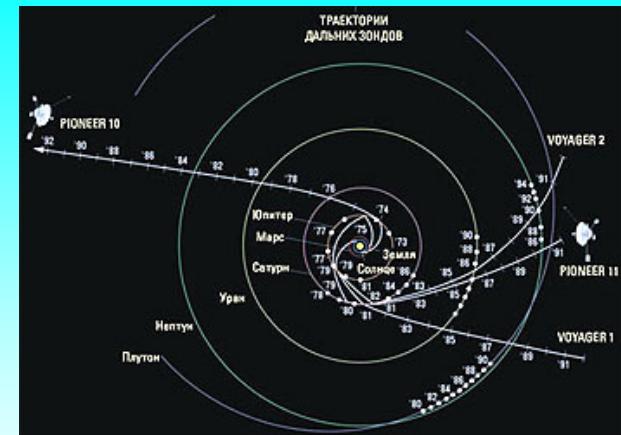




1977

Voyager spacecrafts mission to Saturn

- Launch (1977)
- Start of Jupiter exploring (1979)
- Start of Saturn exploring (1980)
- Voyager mission is still valid (2007)



Phase Separation in Giant Planets:

Jonathan J. FORTNEY, William B. HUBBARD
Icarus, 164 (1) 2003

Atmospheric elemental abundances in Jupiter and Saturn
(mass fractions)

Element	SOLAR	JUPITER <i>Galileo</i>	SATURN <i>Voyager</i>	SATURN revised
H	0.736	0.742	0.92	0.76
He	0.249	0.231 ± 0.04	0.06 ± 0.05	0.215 ± 0.035

1977

Stevenson & Salpeter, *Astrophysical Journal, Suppl.* 35 (1977)

- *The phase diagram and transport properties for hydrogen-helium fluid planets* (p.221-237)
- *The dynamics and helium distribution in hydrogen-helium fluid planets* (p.239-261)

1977

Iosilevskiy I., PhD Thesis, "Non-congruent" condensation in $H_2 + Li$ plasma (1977)

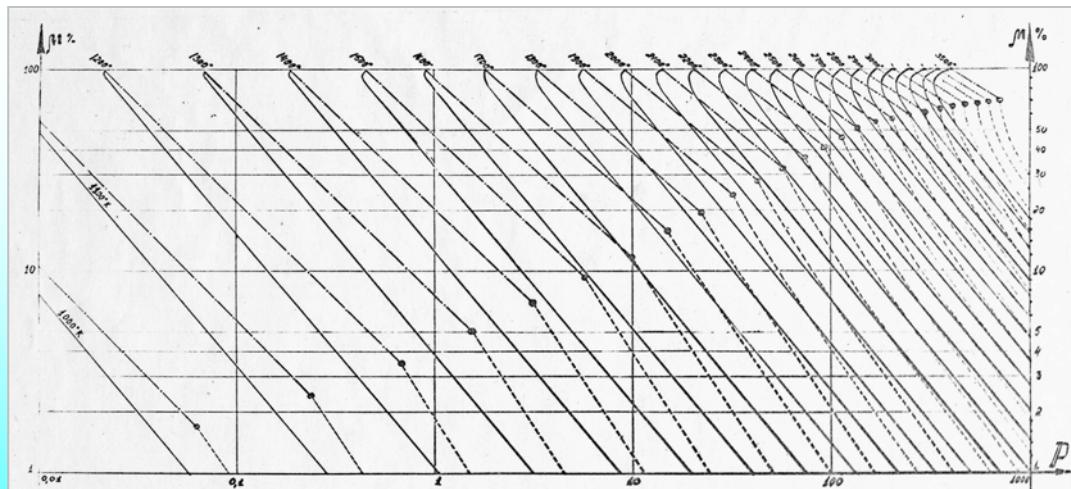


Рис. 37. Граница конденсации в смеси $H_2 + Li$. По оси ординат - мольная доля Li в смеси. Штриховые линии - выпадение в осадок гидрида лития; штрих-пунктирные - выпадение в осадок LiH ; сплошные линии - осадок считается идеальным раствором из Li и LiH .

1989

Galileo spacecraft mission to Jupiter

- Launch (1989)
- Start of Jupiter exploring (1995)
- The end of Galileo mission (Sept. 21, 2003)



Phase Separation in Giant Planets:

Jonathan J. FORTNEY, William B. HUBBARD
Icarus, 164 (1) 2003

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Strongly Coupled Plasma Physics

Int. Conference, Rochester // Chair: Hugh Van Horn

2. DENSE MULTI-IONIC MATERIALS

MICROSCOPIC CORRELATIONS AND PHASE DIAGRAMS FOR DENSE MULTI-IONIC PLASMAS	53
S. Ogata, H. Iyetomi, S. Ichimaru, and H. M. Van Horn	
CRYSTALLIZATION OF DENSE BINARY IONIC MIXTURES. APPLICATION TO WHITE DWARFS COOLING THEORY	63
G. Chabrier and L. Segretain	
ELECTRONIC STRUCTURE AND ENTHALPY OF HYDROGEN AND HELIUM MIXTURES	73
M. Ross, J. E. Klepeis, K. J. Schafer, and T. W. Barbee III	
PLASMA PHASE TRANSITION IN HYDROGEN-HELUM PLASMAS	77
M. Schlange, M. Bonitz, and T. Tschitschjan	

4. ASTROPHYSICS I: GIANT PLANETS, BROWN DWARFS, AND THE SUN

GIANT PLANETS AND THE PLASMA PHASE TRANSITION OF HYDROGEN	111
D. Saumon, G. Chabrier, W. B. Hubbard, and J. I. Lunine	
SEISMOLOGY OF JOVIAN PLANETS AND BROWN DWARFS	121
M. S. Marley	
STRONGLY COUPLED PLASMAS IN URANUS AND NEPTUNE	131
W. B. Hubbard	
RECENT THEORETICAL RESULTS ON BROWN DWARF PROPERTIES AND EVOLUTION	137
J. I. Lunine, D. Saumon, W. B. Hubbard, and A. S. Burrows	
THE HEAVY-ELEMENT CONTRIBUTION TO THE EQUATION OF STATE OF THE SOLAR INTERIOR: THE DIAGNOSTIC POTENTIAL OF HELIOSEISMOLOGY	147
W. Däppen	
ISOCHORES OF A QUANTUM PLASMA NEAR THE FULLY IONIZED LIMIT. APPLICATION TO THE SUN	151
A. Perez, G. Chabrier, and A. Alastuey	

5. ASTROPHYSICS II: DEGENERATE STARS

KINETIC PROPERTIES OF NEUTRON STAR CORES	157
D. G. Yakovlev	
REDUCTION OF DIRECT URCA PROCESS BY NUCLEON SUPERFLUIDITY IN NEUTRON STAR CORES	167
K. P. Levenfish and D. G. Yakovlev	
TOPICS ON THE EQUATION OF STATE IN METALS AND COLD STARS	171
D. Lai and S. L. Shapiro	

Chabrier, Saumon, Hubbard & Lunine
Ap.J 391 (1992)

*The molecular-metallic transition of hydrogen and
the structure of Jupiter and Saturn*

= <> =

Saumon, Hubbard, Chabrier & Van Horn
Ap.J 391 (1992)

*The role of the molecular-metallic transition of
hydrogen in the evolution of Jupiter, Saturn, and
brown dwarfs*

Expected presence of «plasma phase transition» in interiors of Jupiter and Saturn

(Chabrier G., Saumon D., Hubbard W., Lunine J. *Astrophys. Journal* **381** (1992) p.817)

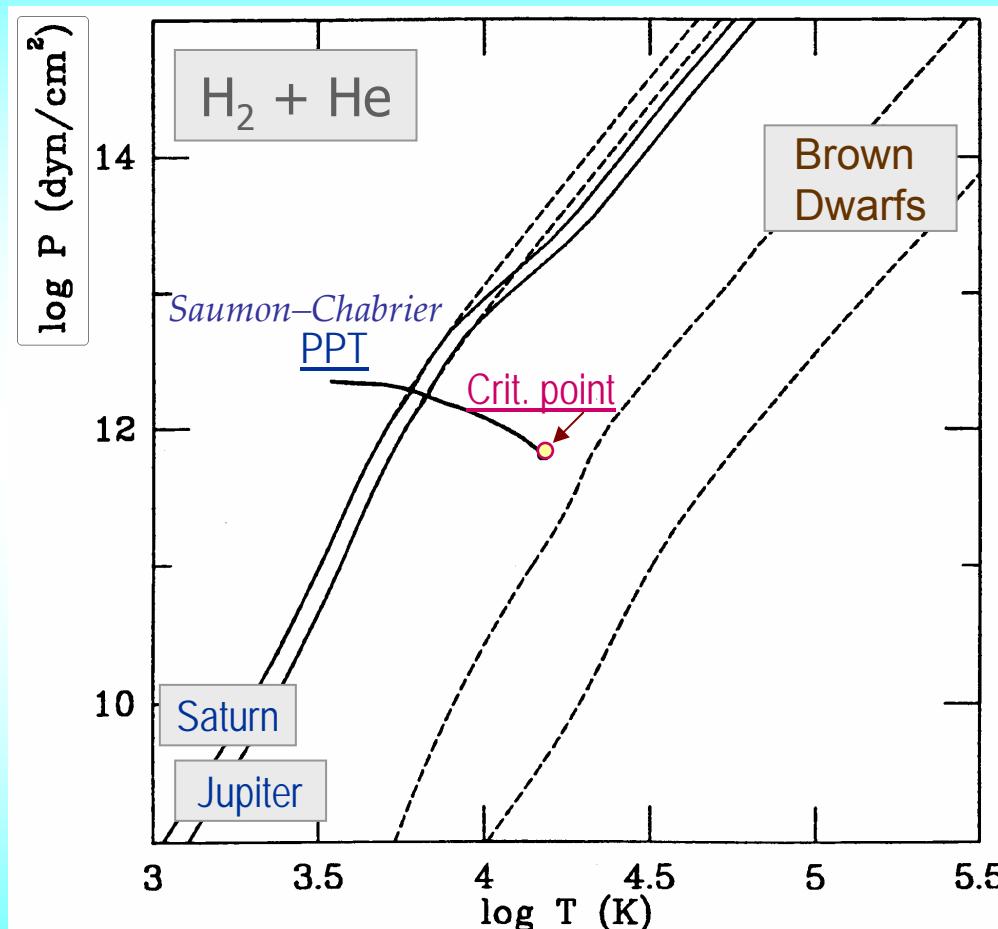


FIG. 3.—Adiabats computed from the EOS described in § 2 with a helium mass fraction $Y = 0.24$. The heavy solid line is the coexistence curve of the plasma phase transition and the critical point is indicated by a dot. Solid lines are computed from the EOS with PPT, dashed lines from the interpolated hydrogen EOS (see text). The temperature of the adiabats at the 1 bar pressure level is, from left to right: 135 (Saturn), 165 (Jupiter), 1500, and 3500 K. In a

Strongly Coupled Plasma Physics

Int. Conference, Rochester // Chair: Hugh Van Horn

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M. S. Marley 121

STRONGLY COUPLED PLASMAS IN URANUS AND NEPTUNE

W. B. Hubbard 131

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W. Däppen 147

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A. Perez, G. Chabrier, and A. Alastuey 151

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TOPICS ON THE EQUATION OF STATE IN METALS AND COLD STARS

D. Lai and S. L. Shapiro 171

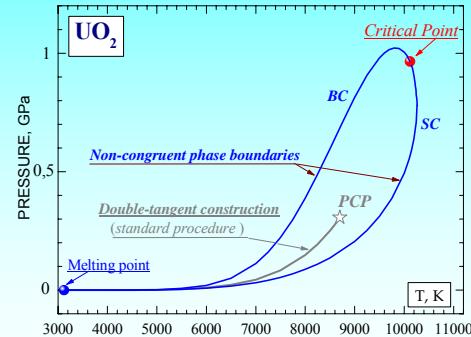
1992

Launch of INTAS-93-66 Project, *Equation of State of Uranium Dioxide up to the Critical Point*

1997

Finish of the first stage of INTAS Project

Conference on *EOS of matter under extreme conditions*, Elbrus



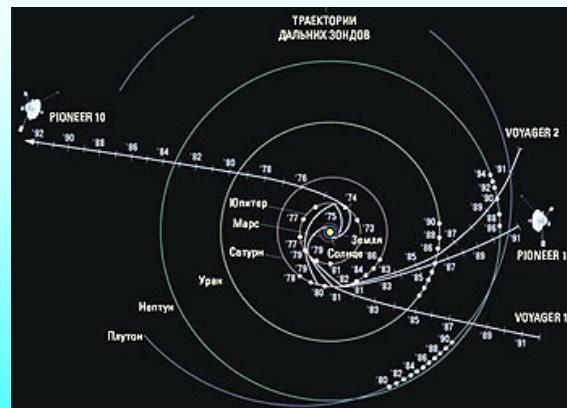
1997 - 2007

Cassini-Huygens mission to Saturn

Launch (Oct. 1997)

Perturbation maneuvers

- Venus-2 (April 1998, June 1999)
- Earth (August 1999)
- Jupiter (December 2000)
- Start of Saturn Exploring (June, 2004)





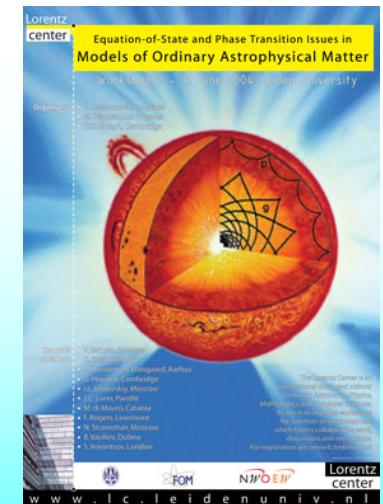
June 2004

Start of Saturn exploring by Cassini-Huygens

June 2004

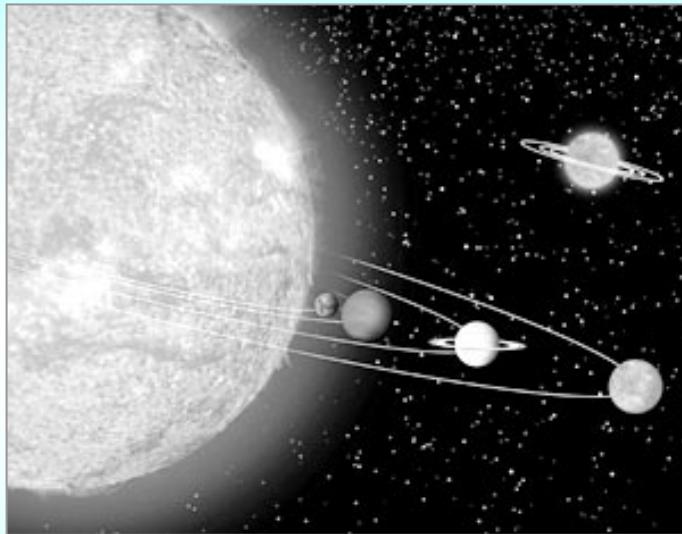
Iosilevskiy I., *Non-congruent phase transitions in astrophysical objects*

International Workshop
Equation-of-State and Phase-Transitions in Ordinary Astrophysical Matter
Leiden, Lorentz Center
(June, 2004)



Dominating hypothesis:

Jupiter & Saturn were formed simultaneously (~ 4,5 GYr)
from the protoplanetary plasma with composition similar to protosolar



Phase Separation in Giant Planets: Inhomogeneous Evolution of Saturn

Jonathan J. FORTNEY, William B. HUBBARD

(*Lunar and Planetary Laboratory, University of Arizona*)

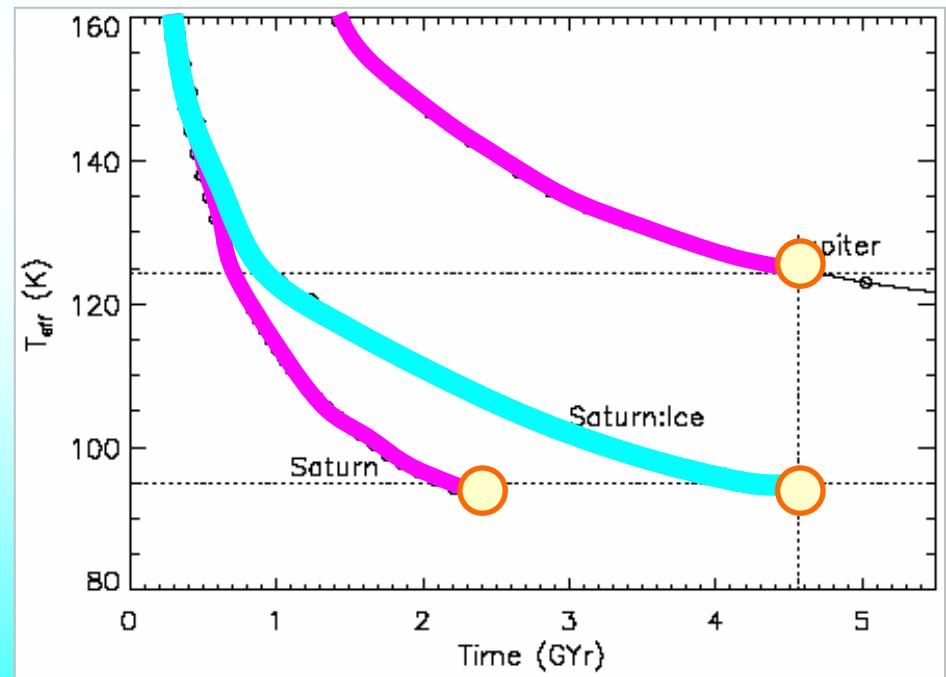
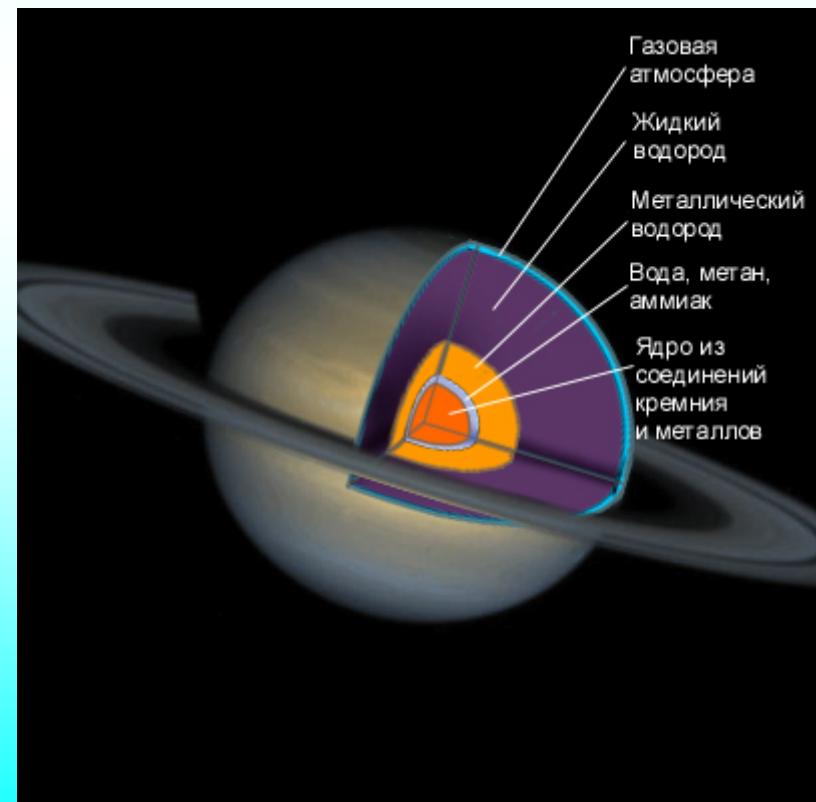
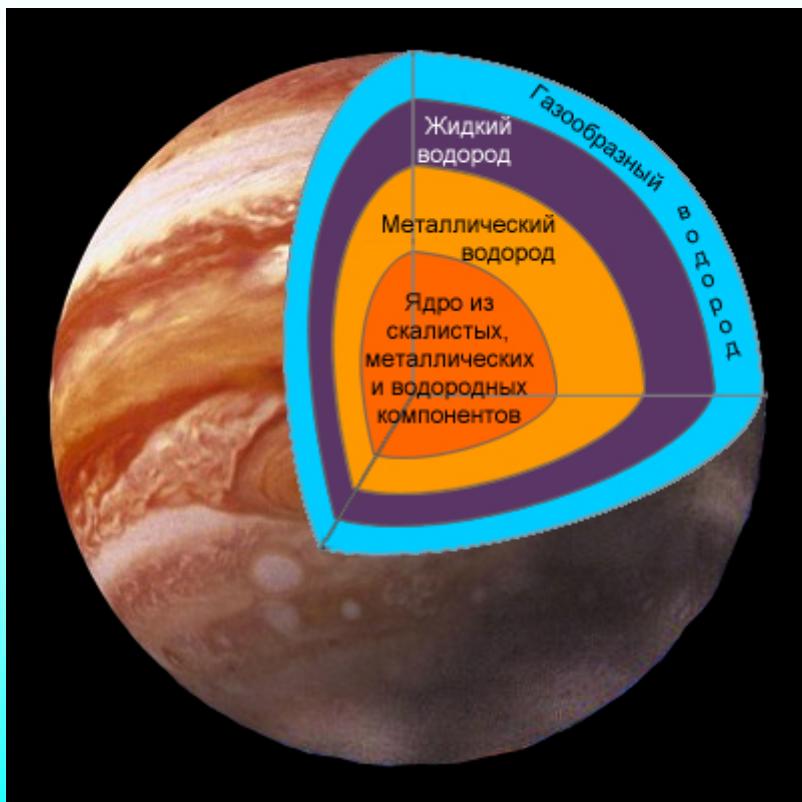
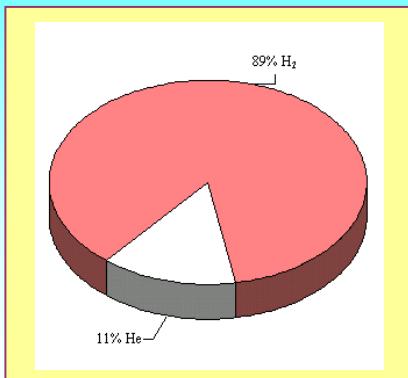


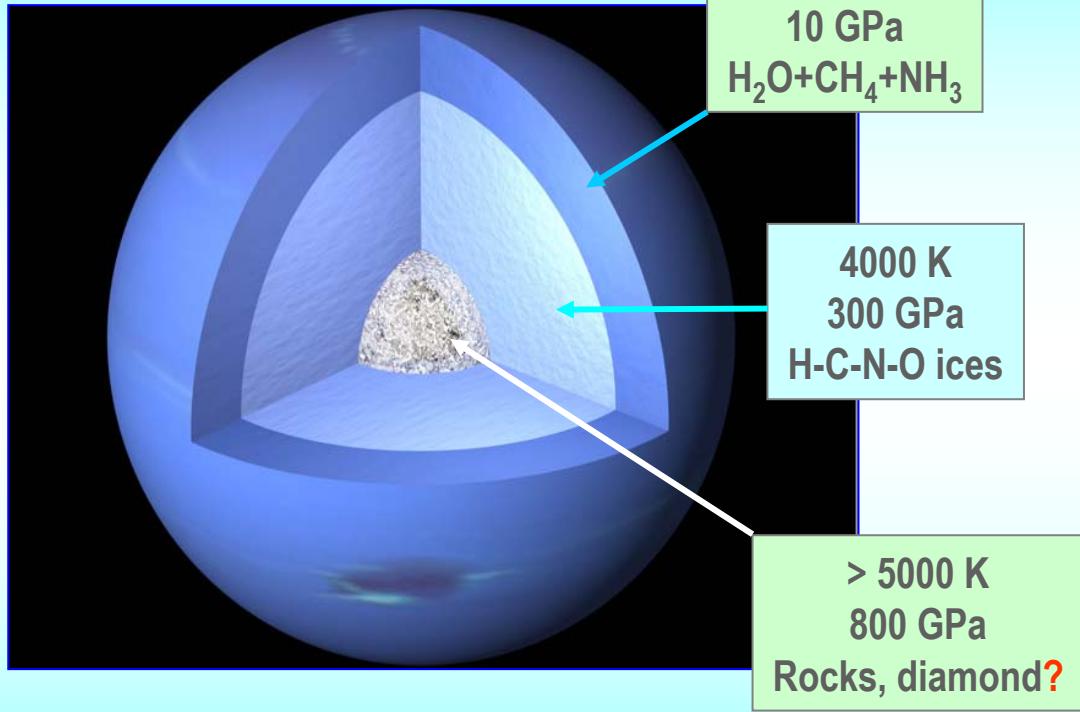
Figure 14. Evolution of Saturn with separation of heavy elements. Homogeneous evolutionary models are labeled "Saturn" and "Jupiter," while the evolution of Saturn with separation of CNO elements is labeled "Saturn:Ice." The

Hypothetical phase transitions in interiors of Jupiter and Saturn

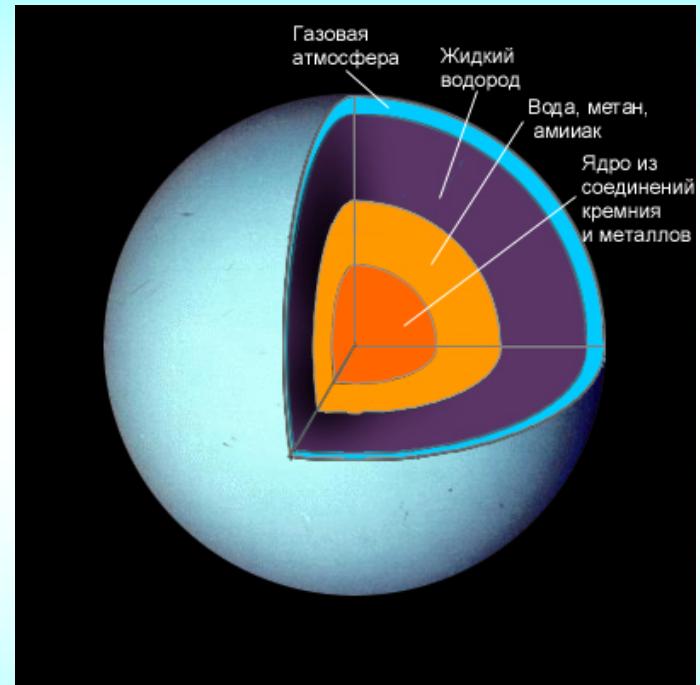


Hypothetical Phase Transitions in interiors of Solar System Giant Planets

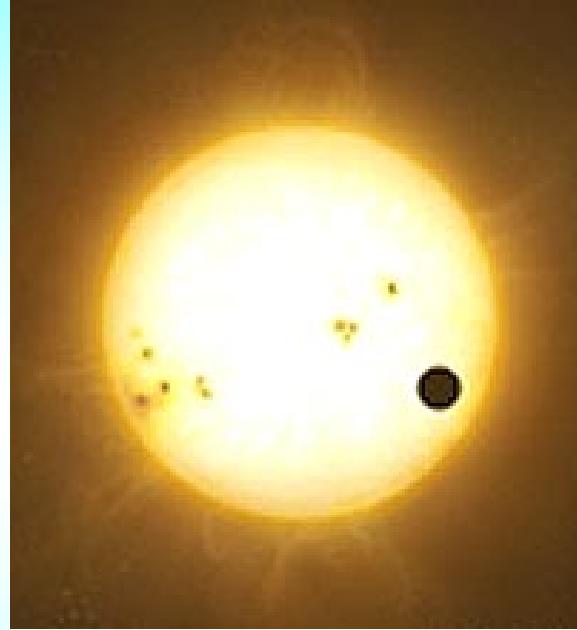
Neptune



Uranus



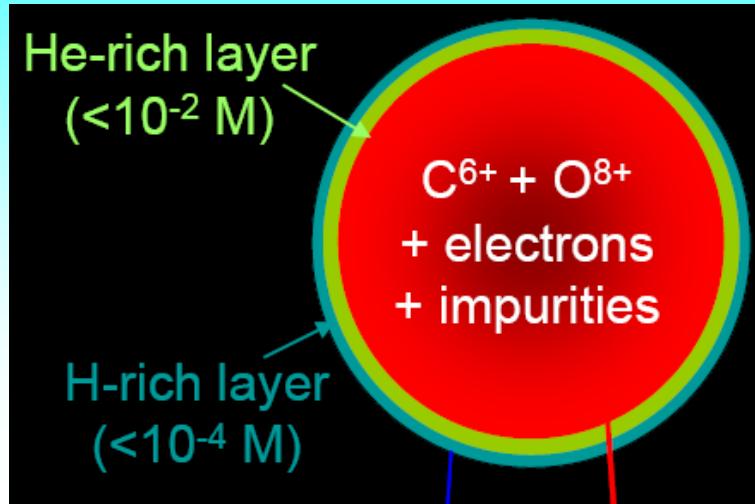
More than 200 «extrasolar» giant-planets have been discovered
(“*Hot Jupiters*”)



EGP-33634

Уже открыто более 200 «внесолнечных» планет—гигантов
(«горячие Юпитеры»)

Hypothetical phase transitions in outer layers of white dwarfs



Jérôme Daligault // SCCS, Moscow, 2005

^{22}Ne



Phase transitions in $H_2 + He$ mixture

“Helium rain”(?)

Hypothetical demixing of He and H_2 ,
in upper and inner layers of Jupiter and Saturn

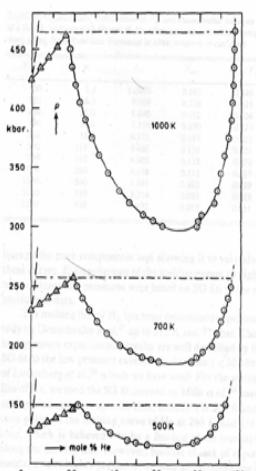
Immiscibility gap in H_2/He mixture

- Low-temperature demixing of *liquid* H_2 and He in *upper* layer of J and S
- High-temperature demixing of *ionized* H and He in *inner* layer of J and S

Phase transitions in H_2/He mixture

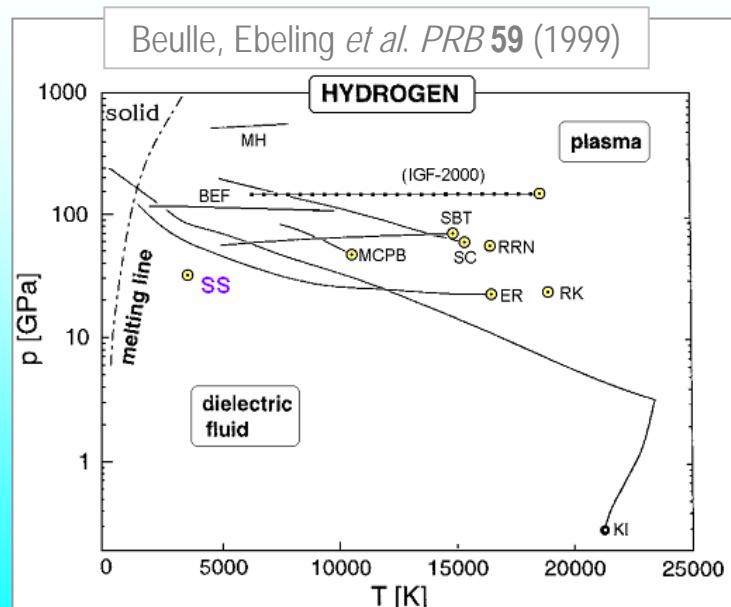
transformed from their **prototypes** in pure hydrogen and helium:

- *hydrogen-like* phase transition(s)
- *helium-like* phase transition(s)



Coexistence of fluid He and H_2
(theory vs. experiment)

Loubeyre P., Toullec R., Pinceaux J. *Phys. Rev. B* **36**, 3723 (1987)
Van den Bergh L., Schouten J. *J. Chem. Phys.* **89** (4) 2336 (1988)
Schouten J., de Kuijper A., Mishel P. *Phys. Rev. B* **44**, 6330 (1991)

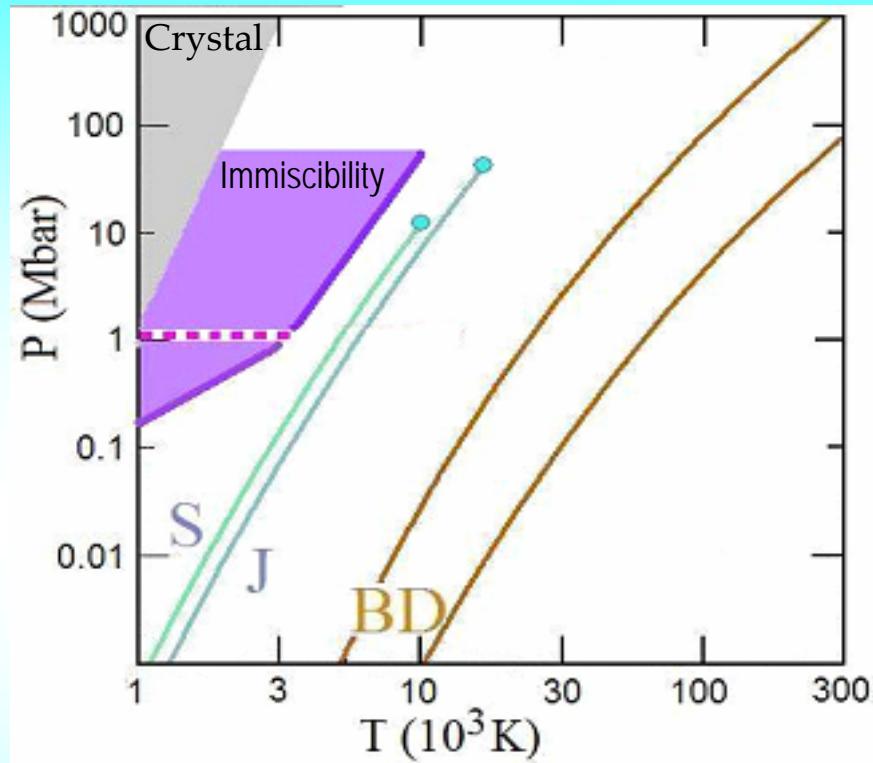
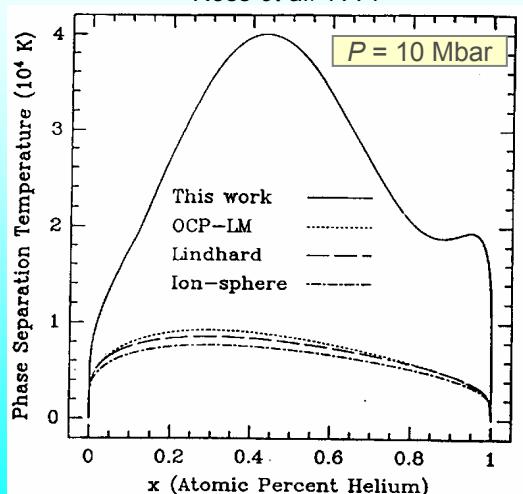
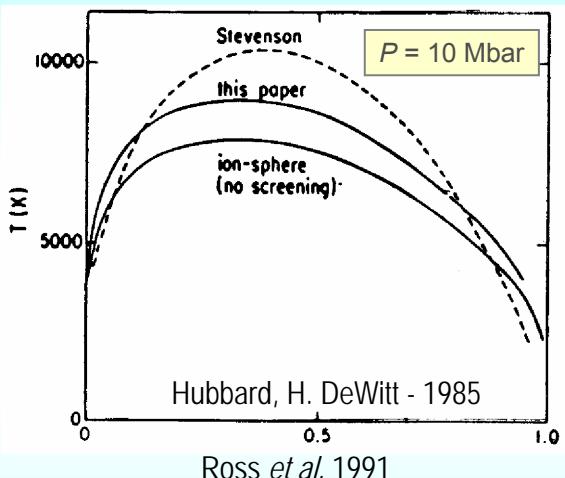




Phase decomposition in H + He mixture

"Helium rain"(?)

Hypothetical demixing of ionized H and He
in inner and upper layers of Jupiter and Saturn



Calculation of demixing boundary for H / He plasma
 $\{ \text{H}^+ + \text{He}^{++} + \text{e}(-) \}$
via Density Functional Theory (DFT) + Mol.Dynamics:

Pfaffenzeller O., Hohl D., Ballone P. *Phys.Rev.Lett.* **74**, 2599 (1995)

- D. Stevenson, E. Salpiter, *Astrophysical Journal* **35**, 221 (1977)
W. Hubbard, H. DeWitt, *Astrophysical Journal*, **290**, 388 (1985)
M. Ross, J. Klepeis, K. Shafer, T. Barbee III - *Science*, **254** 986 (1991)
SCCS Conference, Rochester, (1993)

Expected presence of «plasma phase transition» in interiors of Jupiter and Saturn

(Chabrier G., Saumon D., Hubbard W., Lunine J. *Astrophys. Journal* **381** (1992) p.817)

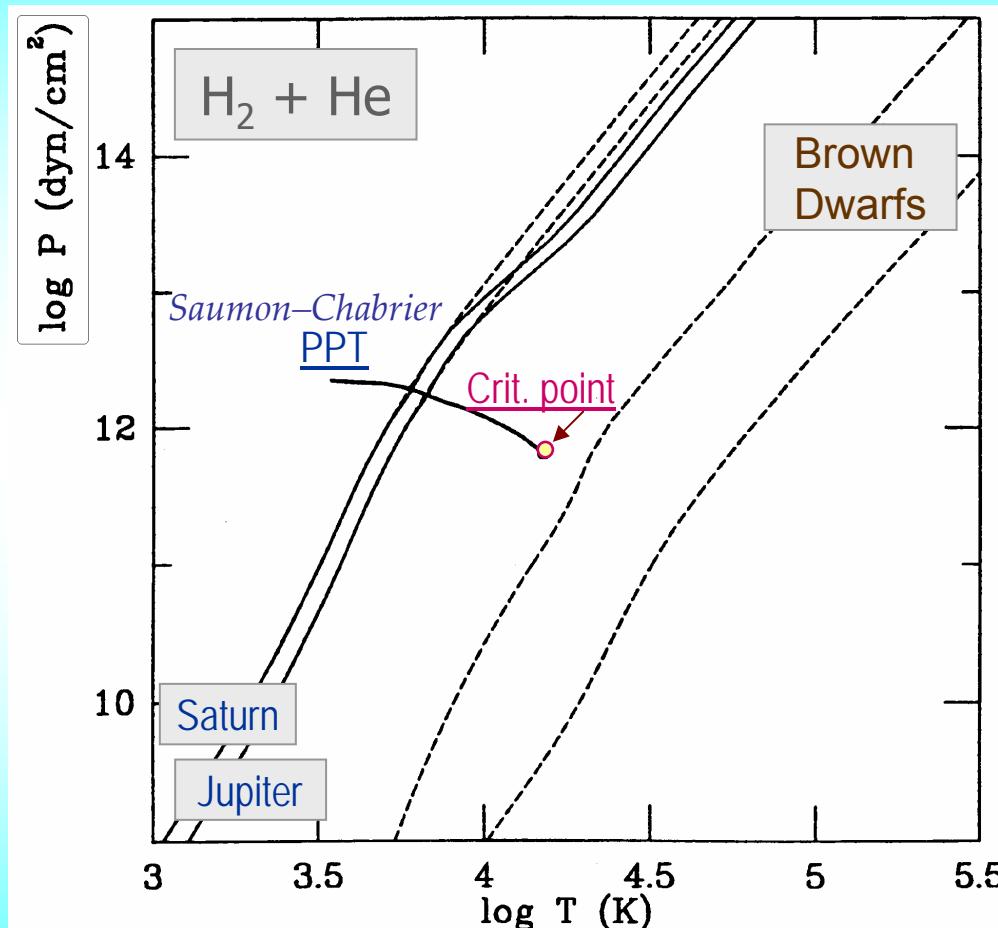


FIG. 3.—Adiabats computed from the EOS described in § 2 with a helium mass fraction $Y = 0.24$. The heavy solid line is the coexistence curve of the plasma phase transition and the critical point is indicated by a dot. Solid lines are computed from the EOS with PPT, dashed lines from the interpolated hydrogen EOS (see text). The temperature of the adiabats at the 1 bar pressure level is, from left to right: 135 (Saturn), 165 (Jupiter), 1500, and 3500 K. In a

Typical picture of plasma phase transition expected in H₂/He mixture in interior of Jupiter and Saturn

Chabrier G., Saumon D., Hubbard W., Lunine J. (SCCS-1992, Rochester)

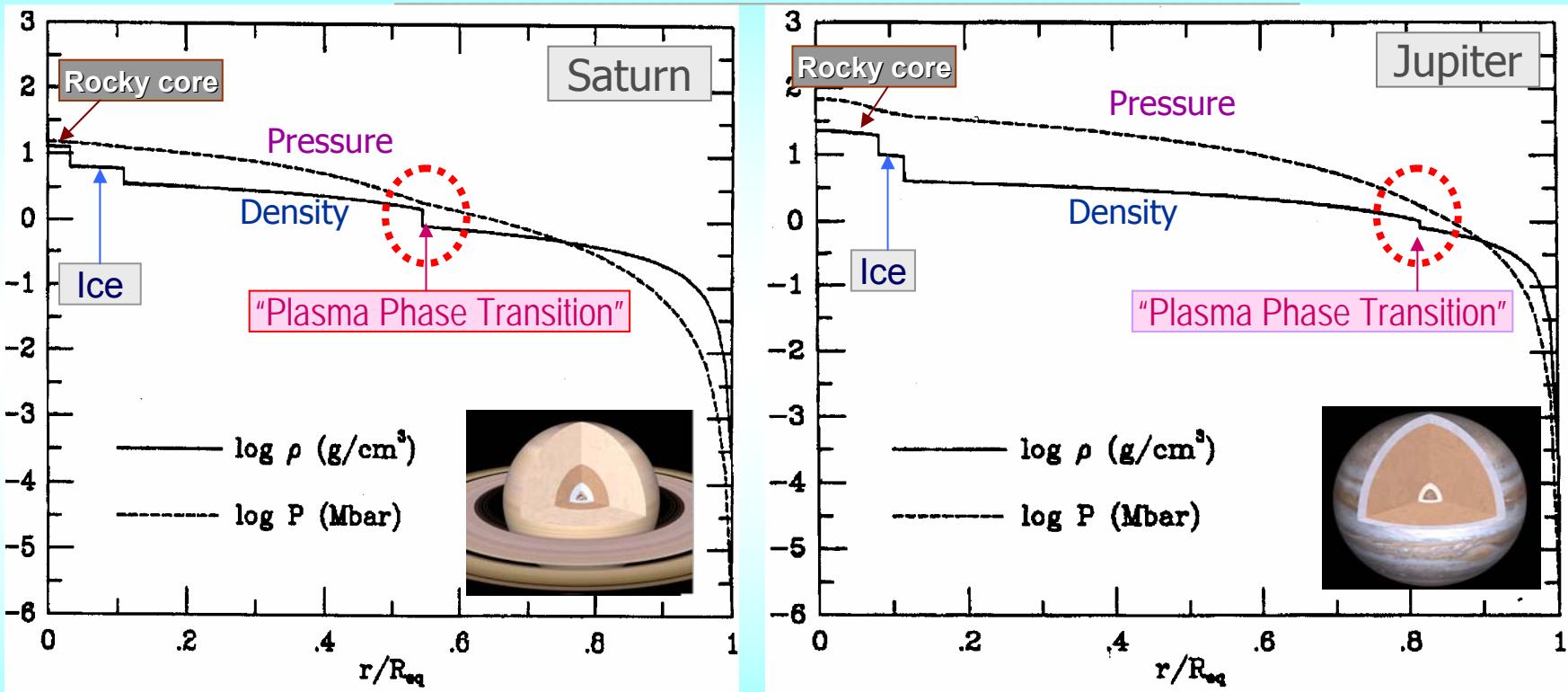


Fig. 1. Pressure and density profiles of optimized models of Jupiter (top panel) and Saturn (bottom panel), plotted as a function of mean radius. Discontinuities in the density clearly mark the boundaries of the four layers of the models: rocky core, ice mantle, metallic and molecular

Typical picture of plasma phase transition expected in interior of Giant Planets

Jupiter

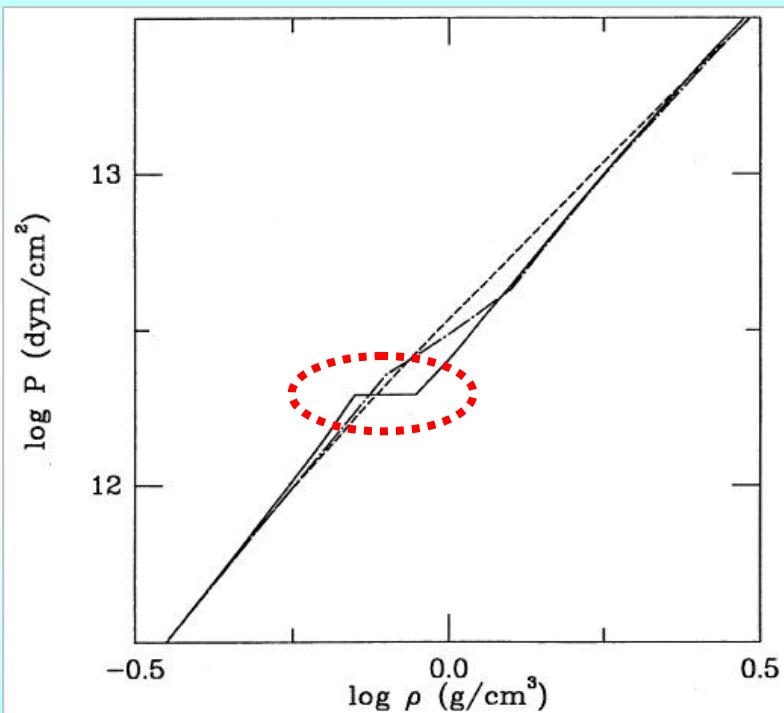


FIG. 2.—Pressure-density relation along a pure hydrogen Jupiter adiabat ($T = 165$ K at $P = 1$ bar). The MH EOS is shown by the long dash-dot curve,

Neptune

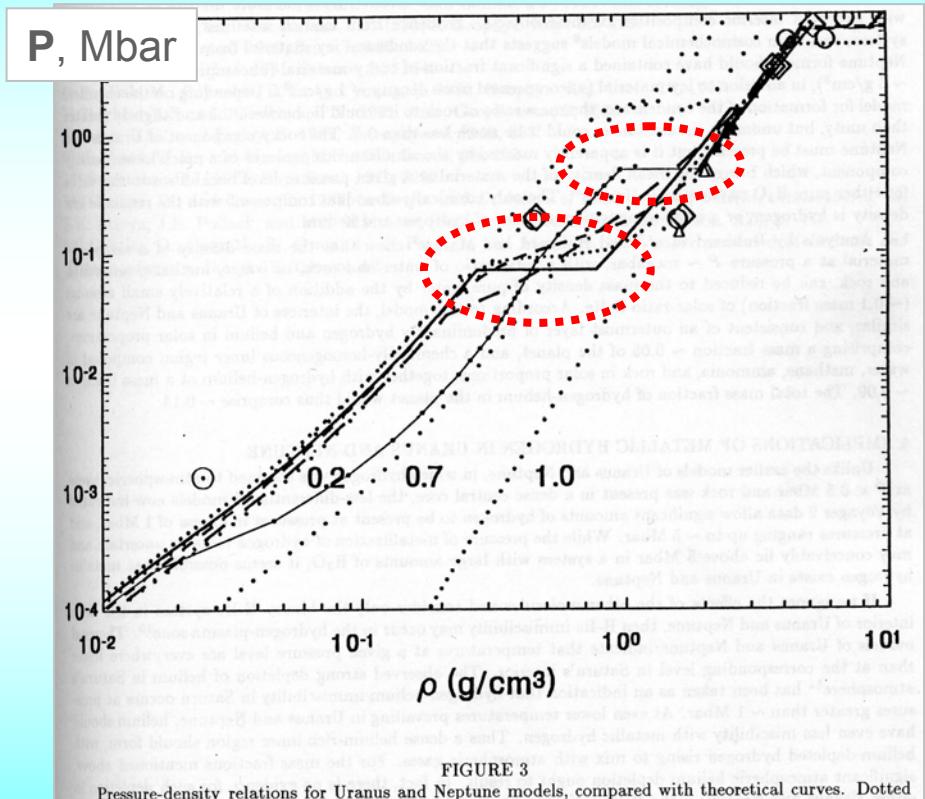
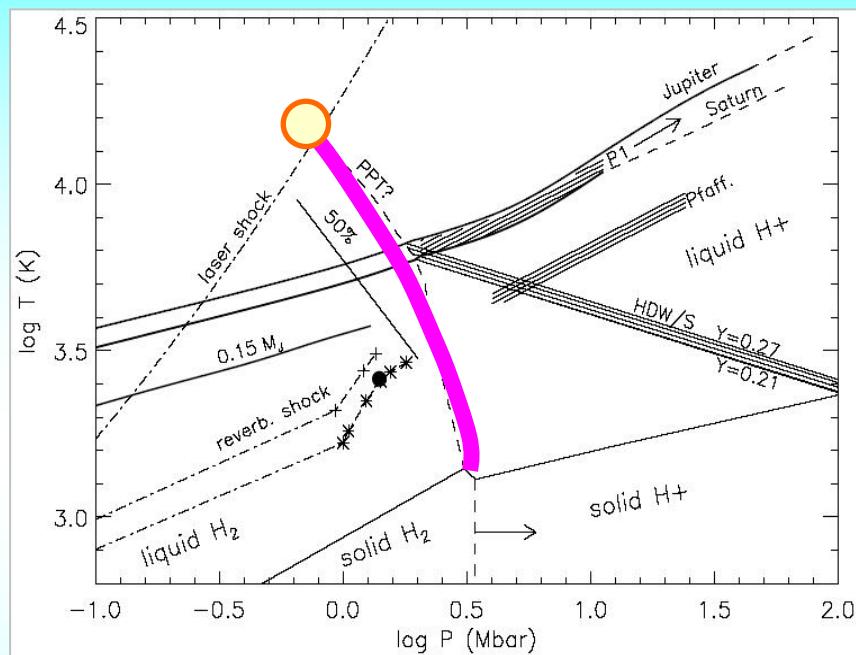
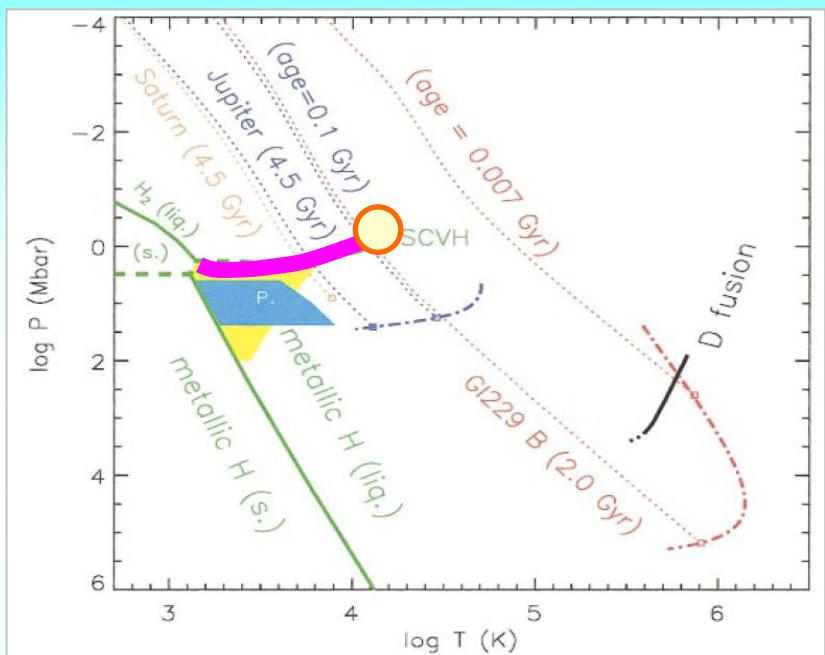


FIGURE 3
Pressure-density relations for Uranus and Neptune models, compared with theoretical curves. Dotted

Typical picture of plasma phase transition expected in interior of Giant Planets



BURROWS, HUBBARD, LUNINE and LIEBERT
The theory of brown dwarfs and extrasolar giant planets
 Rev. Mod. Phys. 73, 719-769, 2001

J. FORTNEY & W. HUBBARD
Phase Separation in Giant Planets
 Icarus, 164, 2003

Hypothetical “Plasma Phase Transition” in hydrogen

Relevant to the problem of Saturn and Jupiter formation

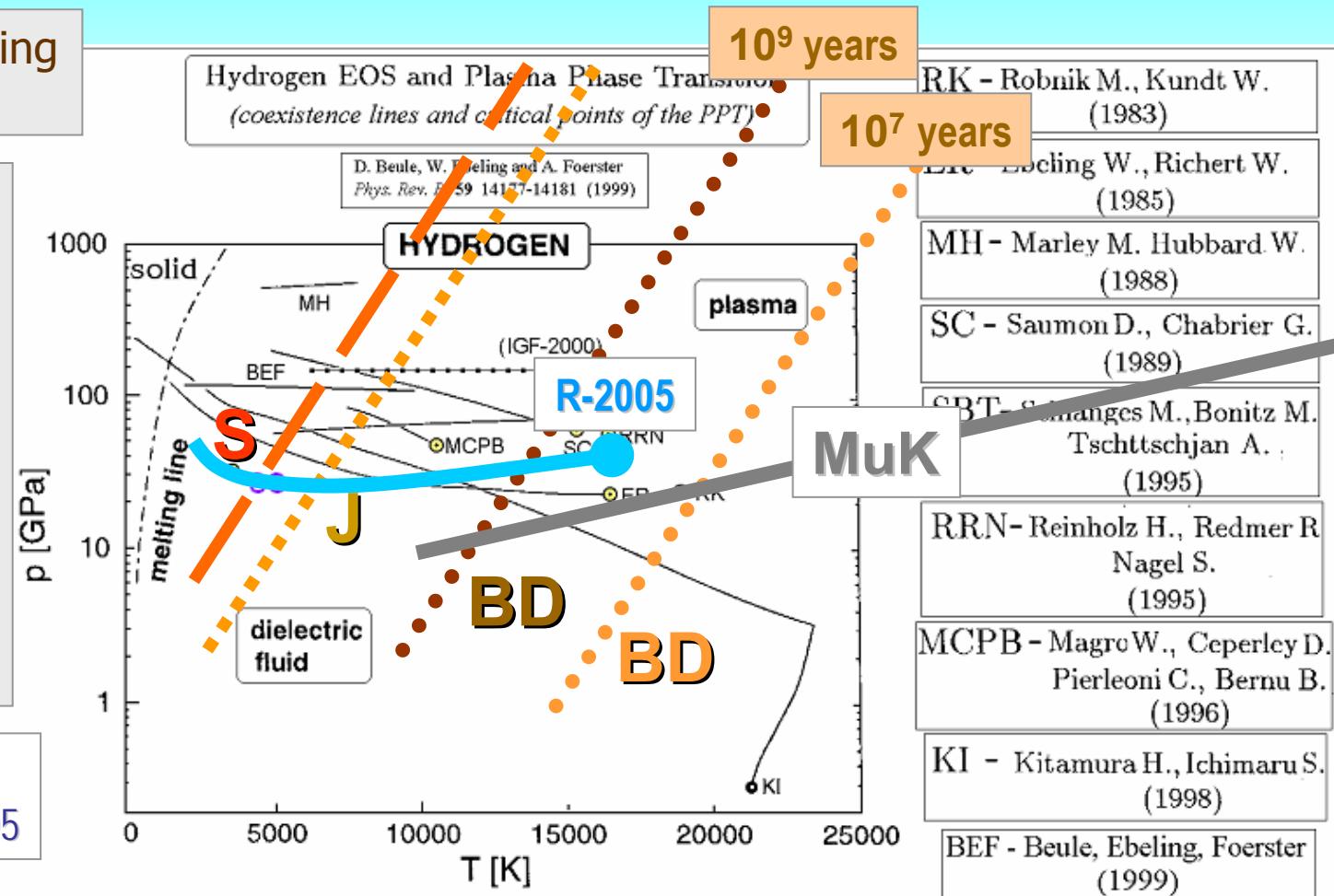
(*) Beulle, Ebeling
et al. (1999)

“MuK”
“Dissociative” PT
Mulenko, Khomkin,
et al. (2001)
(chemical picture)

Good agreement with
ab initio calculations
Flinov, Levashov
Bonitz, Fortov. (2001)
(Quantum Monte-Carlo)
Phys. Lett. A 289 (2001)

R-2005
Redmer et al. 2005

S – Saturn
J – Jupiter
BD – Brown Dwarfs

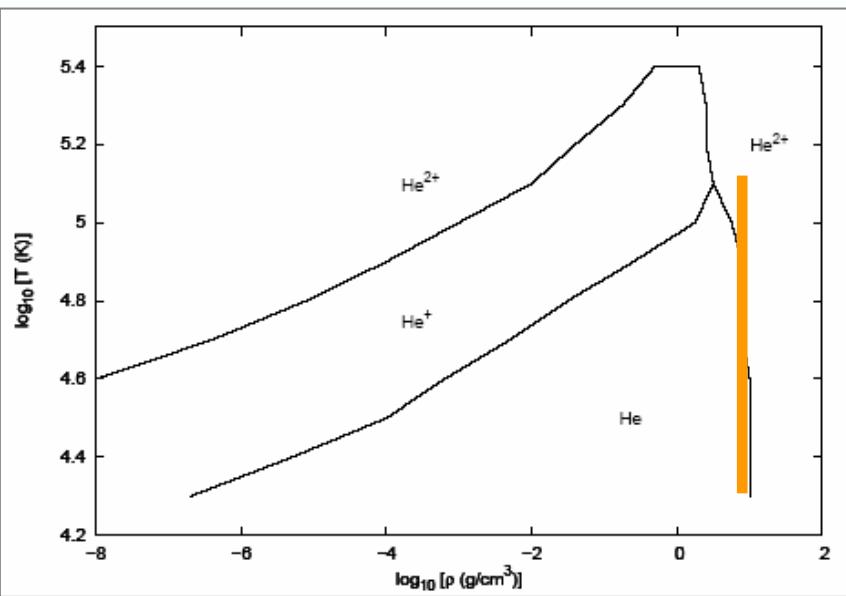


SS - Stevenson D, Salpeter E. (1977) ApJ. Sp.35,221

Hypothetical Plasma Phase Transition (transitions) in helium

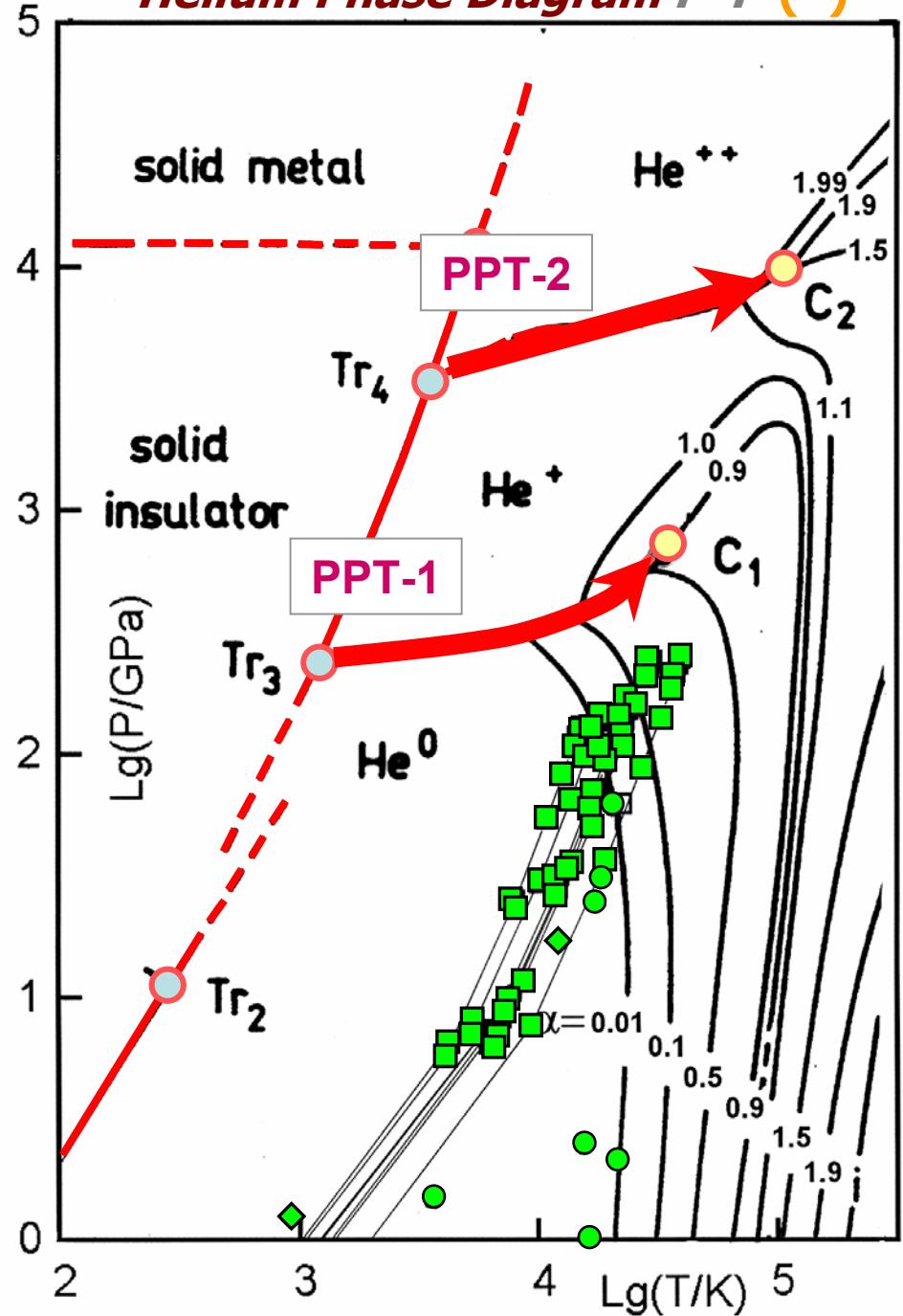
Ebeling, Foerster *et al.* (1991)

PPT?



Winisdoerffer Ch., Chabrier G.,
Free-energy model for fluid helium at high density
Phys.Rev.E (2005)

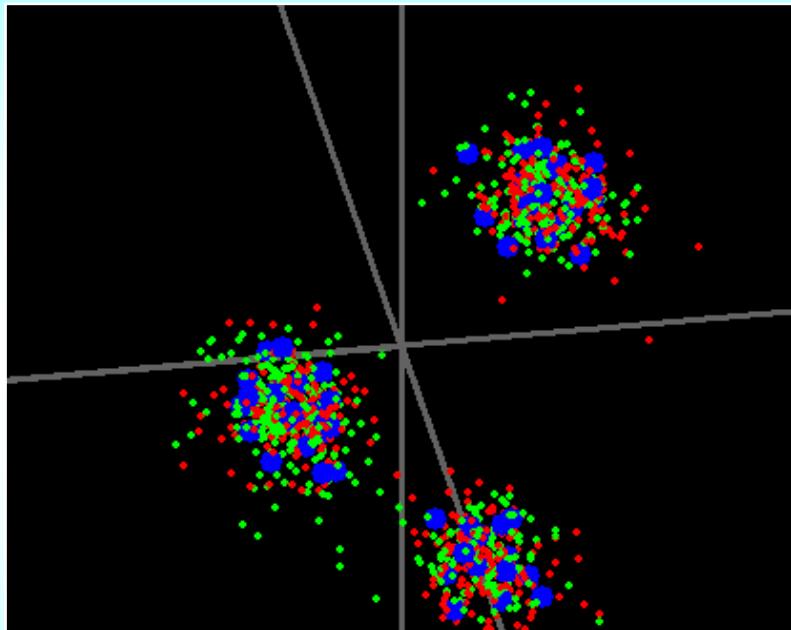
Helium Phase Diagram P-T ()*



Quantum Monte-Carlo Simulations

(V. Filinov, M. Bonits, P. Levashov, V. Fortov)

“Plasma” phase transition in hydrogen



$T = 10000 \text{ K}$, $n = 3 \cdot 10^{22} \text{ cm}^{-3}$

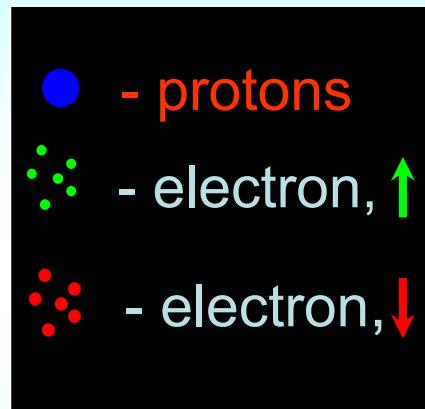


Figure from:
V. Filinov,
P. Levashov
et al.
SCCS-2005,
Moscow

Theoretical prediction of "dissociative" fluid-fluid phase transition in liquid hydrogen (deuterium)

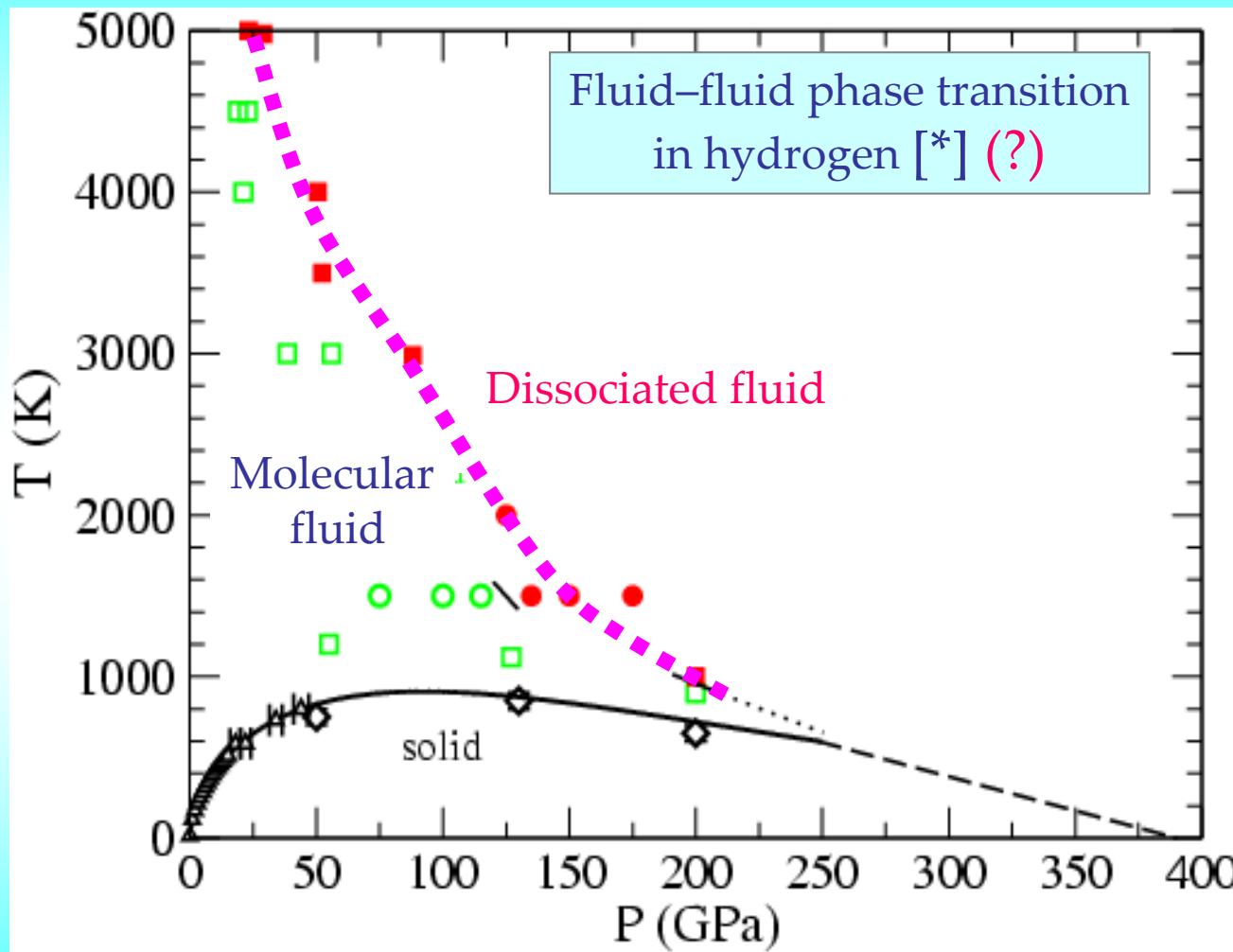


Figure from:
Giulia Galli,
SCCS-2005,
Moscow



Density Functional Theory + Molecular Dynamics

[*] Scandolo S. *PNAS* 100, 3051 (2003) // Bonev S., Militzer B., Galli G. *PRB* (2004)

Theoretical prediction of "dissociative" fluid-fluid phase transition in liquid hydrogen

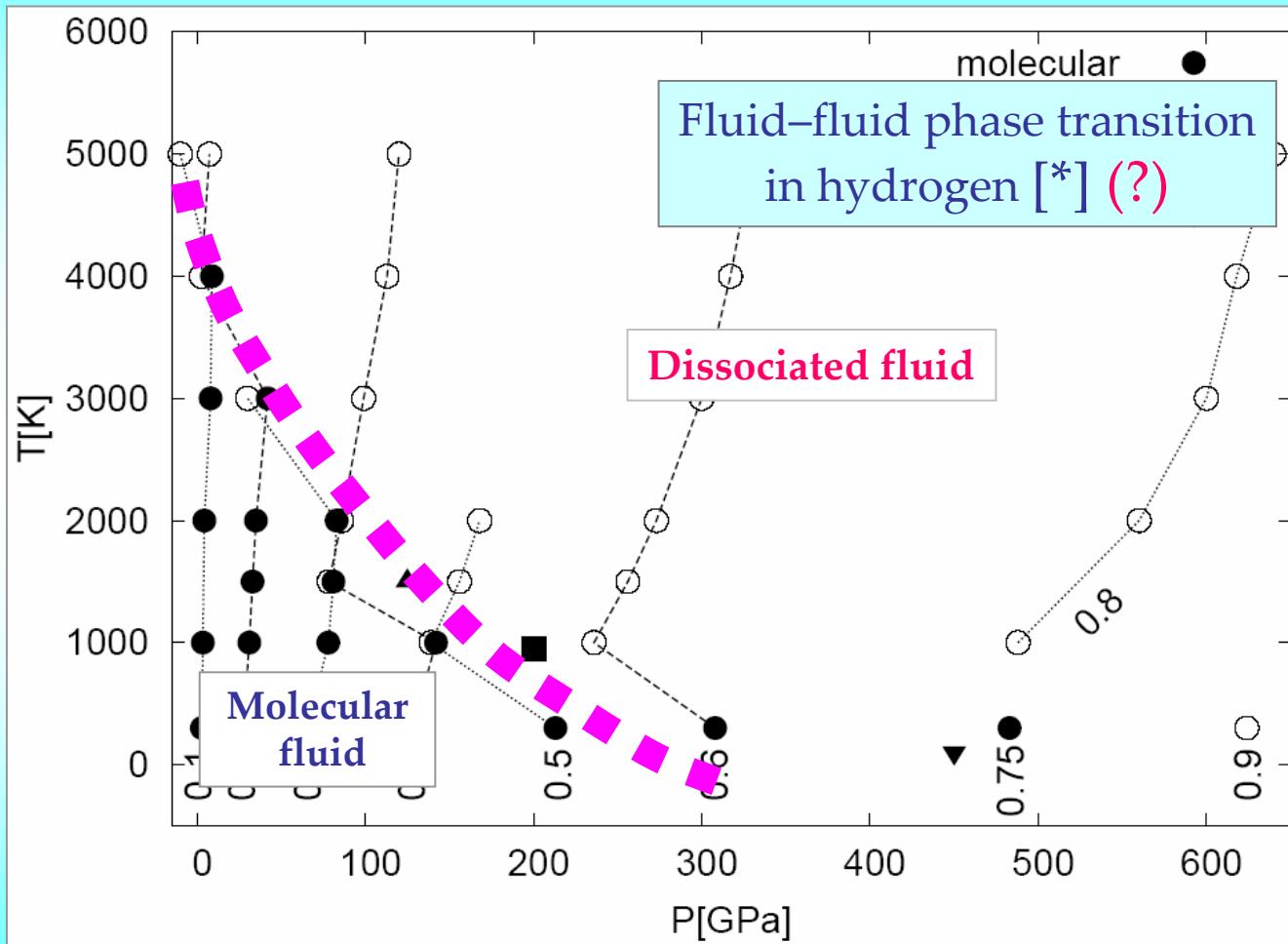


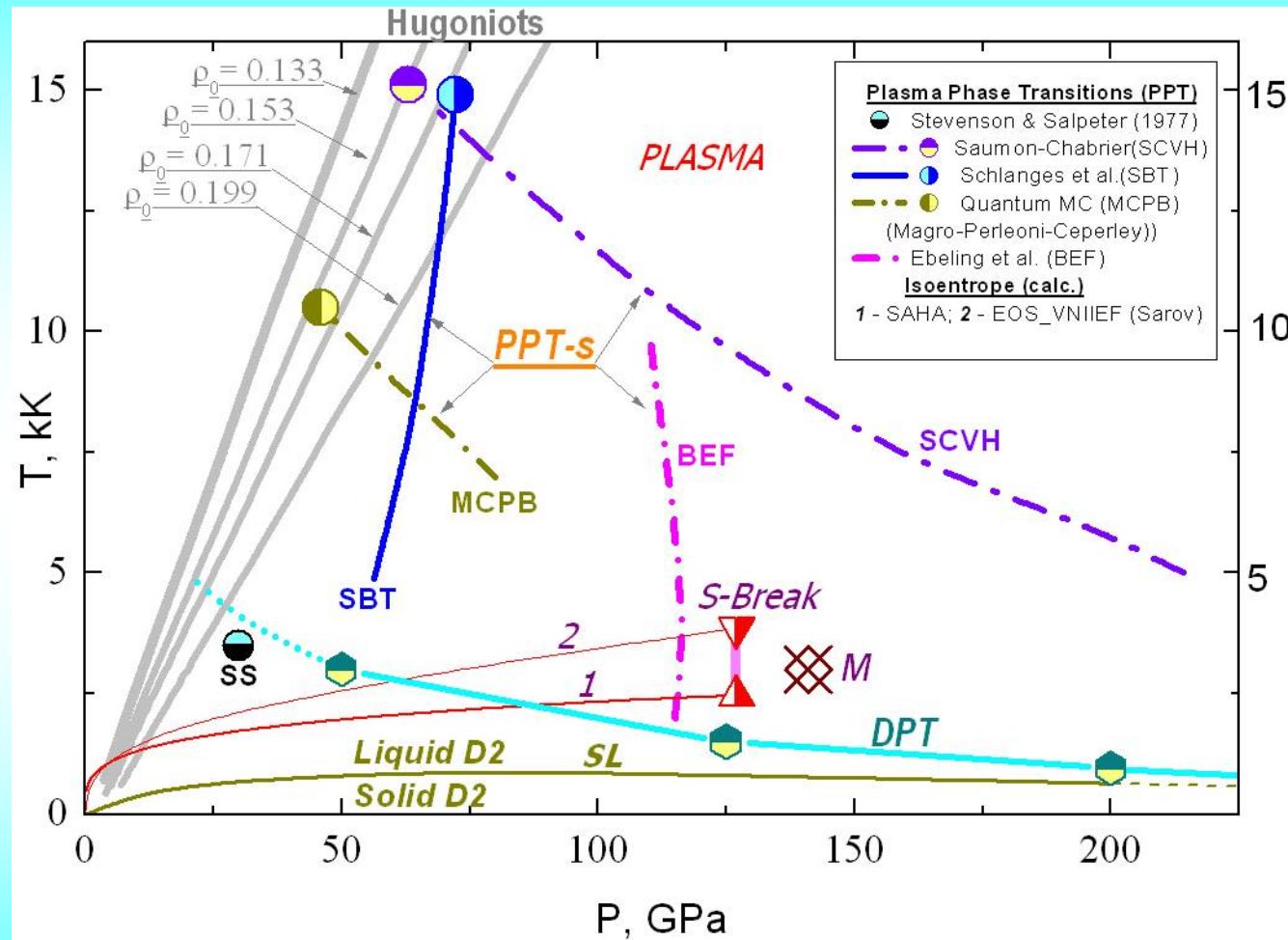
Figure from:
C. Toepffer et al,
"Physics of
High Energy Density
in Matter"
(2007)



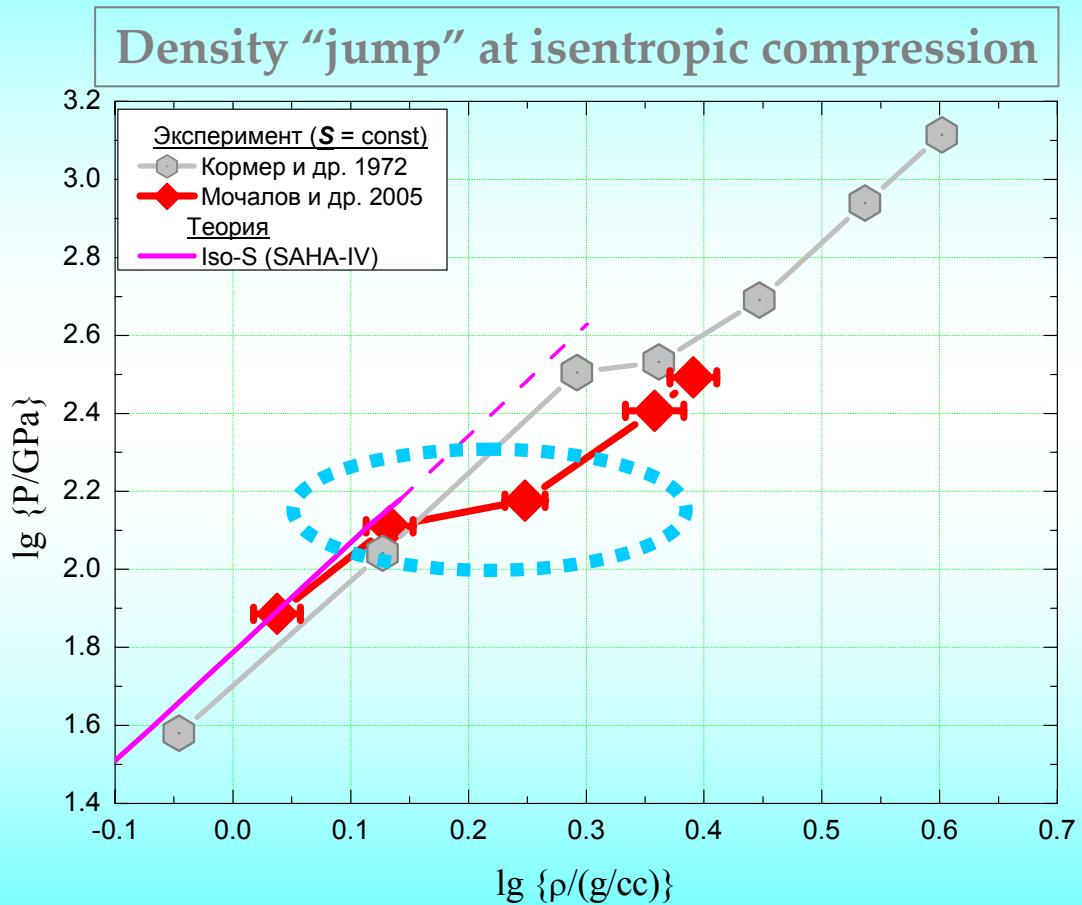
Wave Packet Molecular Dynamics (WPMD) Theory (2007)

[*] Jakob B. Doctoral Thesis, Erlangen University (2006) // Submitted to PRE (2007)

Problem of experimental confirmation of theoretically predicted phase transitions in hydrogen



Quasi-isoentropic compression gaseous deuterium up to the pressure 75-300 GPa

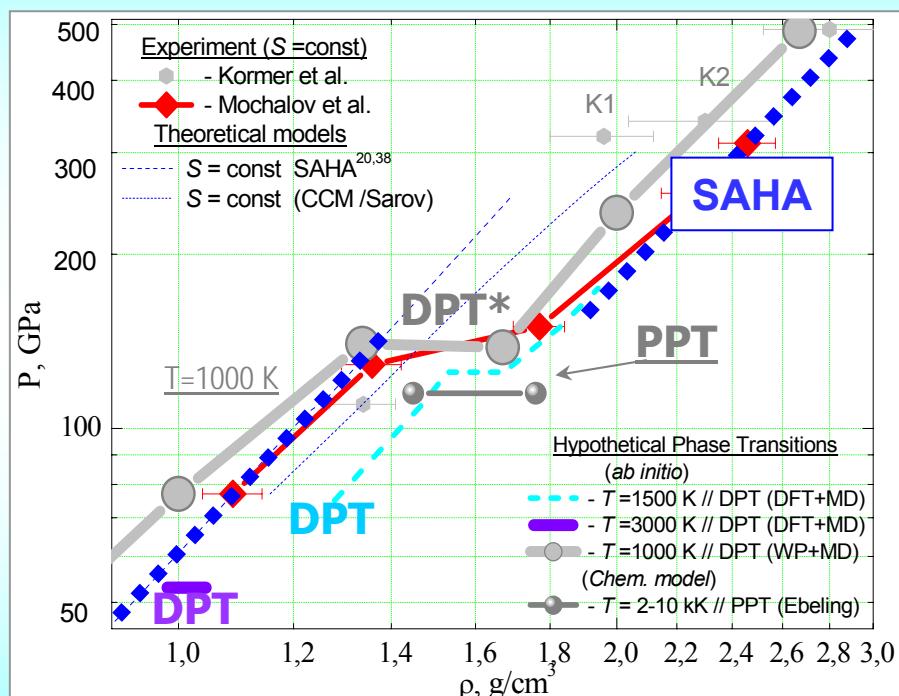


After:
**M. Mochalov
*et al.***
 SCCS-2005,
 Moscow

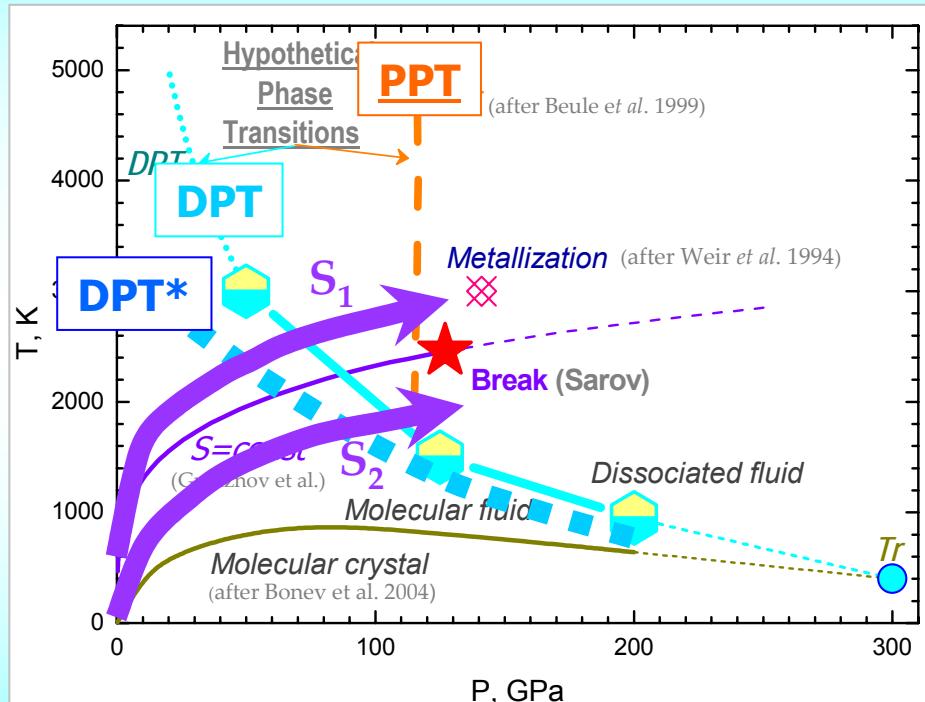
SAHA-IV: - Chemical model (Gryaznov et al.) with modified Coulomb corrections and interaction parameters
 $\text{H}_2\text{-H}_2 / \text{H}_2\text{-H} / \text{H-H}$ – in accordance with “atom-atomic” approximation (E. Yakub – *Physica B* **265** 31 (1999))

Density break in isentropic compression of gaseous deuterium \leftrightarrow hypothetical phase transition (?)

Pressure-Density Diagram



Pressure-Temperature Diagram



PPT – “Ionization driven” phase transition (Beule D., Ebeling W. et al. PRB, 1999)

DPT – “Dissociation driven” phase transition (Scandolo S. 2003 // Bonev S. Militzer B. Galli G. 2004)

DPT* – “Dissociation driven” phase transition (Ab Initio: WPMD // Jakob et al. 2007)



Hypothetical phase transition in H_2/He mixture \leftrightarrow \leftrightarrow Planets evolution problem (?)

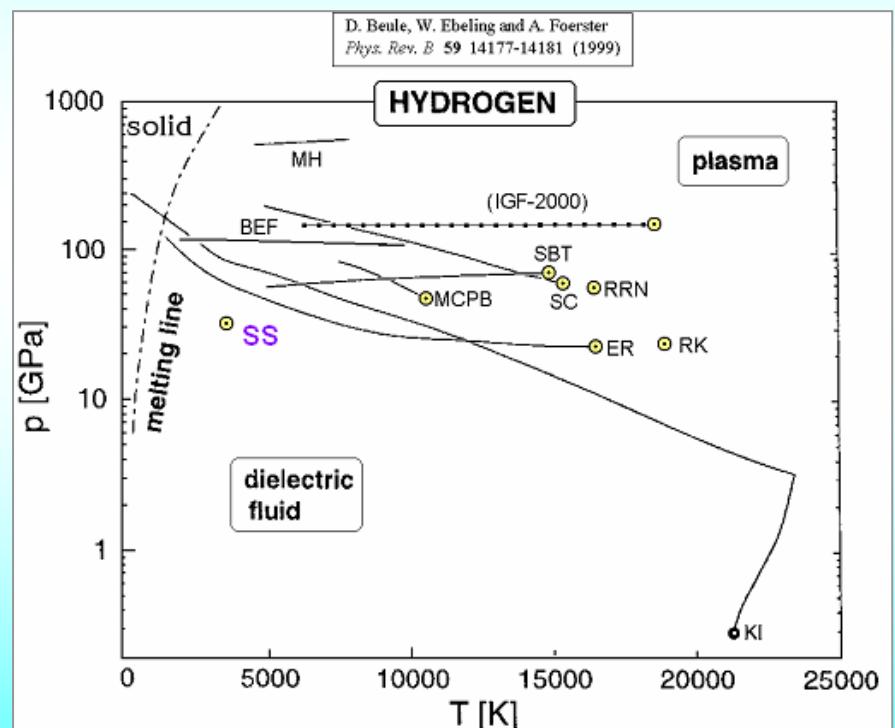
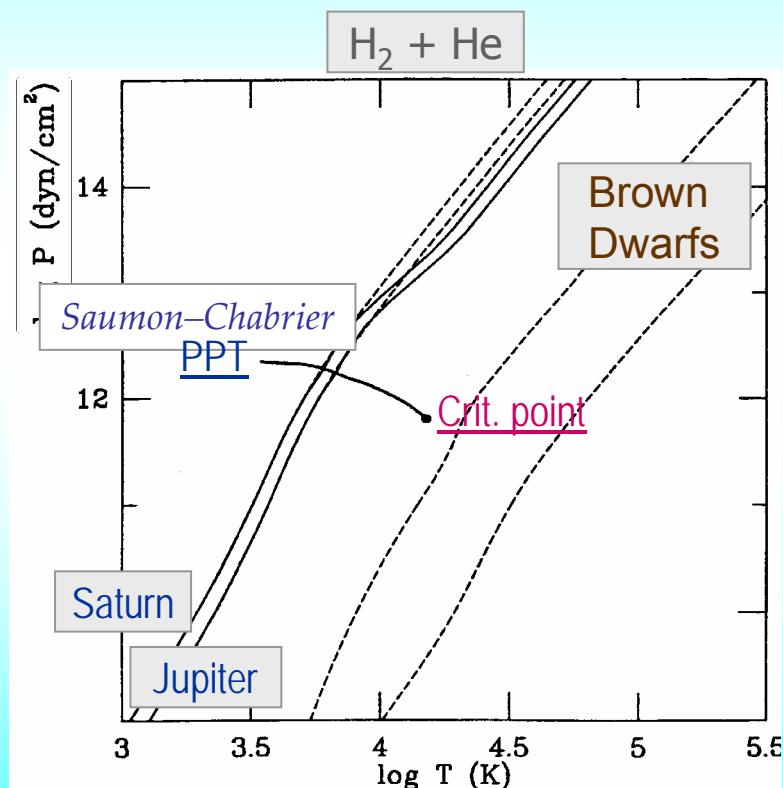
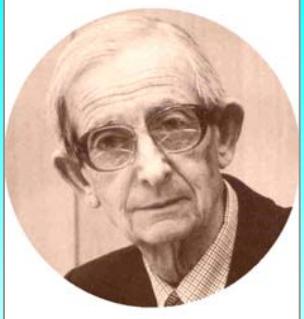


FIG. 3.—Adiabats computed from the EOS described in § 2 with a helium mass fraction $Y = 0.24$. The heavy solid line is the coexistence curve of the plasma phase transition and the critical point is indicated by a dot. Solid lines are computed from the EOS with PPT, dashed lines from the interpolated hydrogen EOS (see text). The temperature of the adiabats at the 1 bar pressure level is, from left to right: 135 (Saturn), 165 (Jupiter), 1500, and 3500 K. In a



Int. Conference "Yu. Khariton's Science Readings"

Extreme State of Matter

Russian Federal Nuclear Center, Sarov, March 2001

Non-congruent phase coexistence in uranium-oxygen plasma

Igor Iosilevskiy

Moscow Institute of Physics and Technology (State University)

Victor Gryaznov & Vladimir Fortov

Institute of Problems of Chemical Physics RAS, Chernogolovka, Russia

Eugene Yakub

Odessa State Economic University, Ukraine

Alexander Semenov

Moscow Power Engineering Institute, Russia

Claudio Ronchi

Institute for Transuranium Elements, JRC, Karlsruhe, Germany

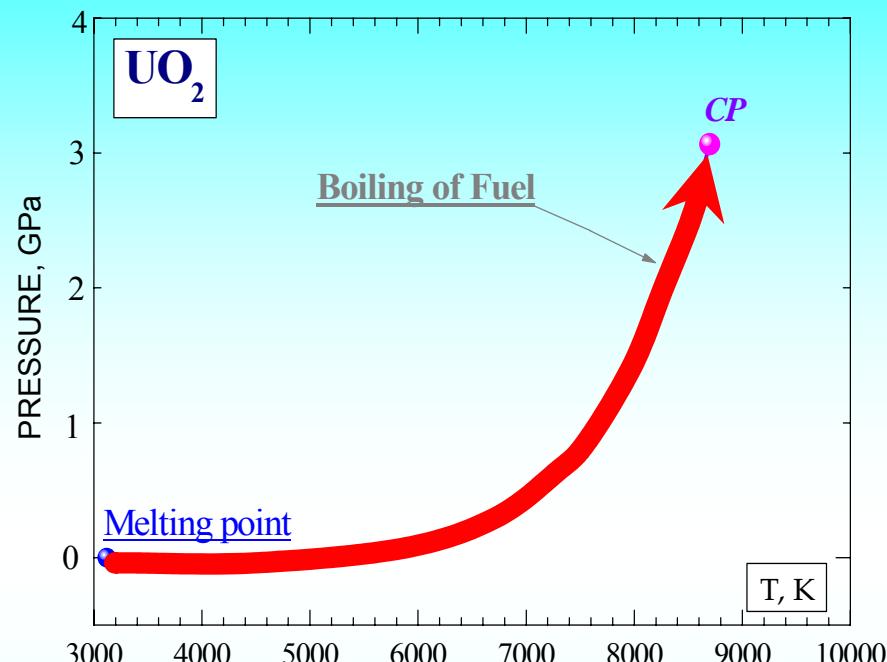
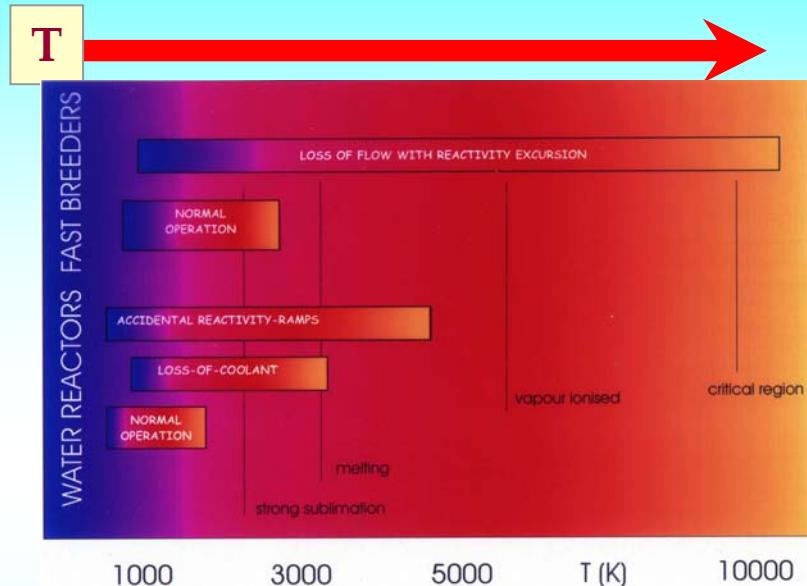
Gerard J. Hyland

University of Warwick, Coventry, United Kingdom



Non-Congruent Phase Transition in Uranium Dioxide

Expected Temperature at Hypothetical **Severe Accident**
at Fast-Breeder **Nuclear Reactor**



INTAS Project (1995–2002)

Cooperation: MIPT – IHED RAS – IPCP RAS – OSEU – MPEI ⇔ ITU (JRC, Germany)

Project Coordinator – C. Ronchi (ITU, JRC) ⇔ Project Supervisor – V. Fortov

ISTC Project (2002–2005)

Cooperation: MIPT – IHED RAS – IPCP RAS – ITEP – VNIIIEF ⇔ GSI (JRC, Germany)

Project Manager – B. Sharkov (ITEP, Moscow) ⇔ Project Science Supervisor – V. Fortov



JOINT RESEARCH CENTRE
Institute for Transuranium Elements



MIPT



OSEU



Non-congruent phase transition in uranium dioxide

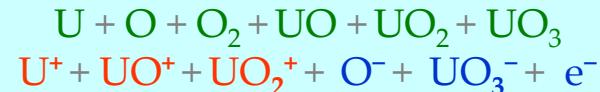
Sketch of theoretical approach

Quasi-chemical representation for liquid & gaseous phases

Ionic Model
(Liquid)



Multi-molecular Model
(Liquid & Gas)



Interactions: (*Pseudopotential components*)

- Intensive Short-range Repulsion
- Coulomb Interaction between Charged Particles
- Short-range Effective Attraction between all Particles

Interaction corrections: (*Modified for mixtures*)

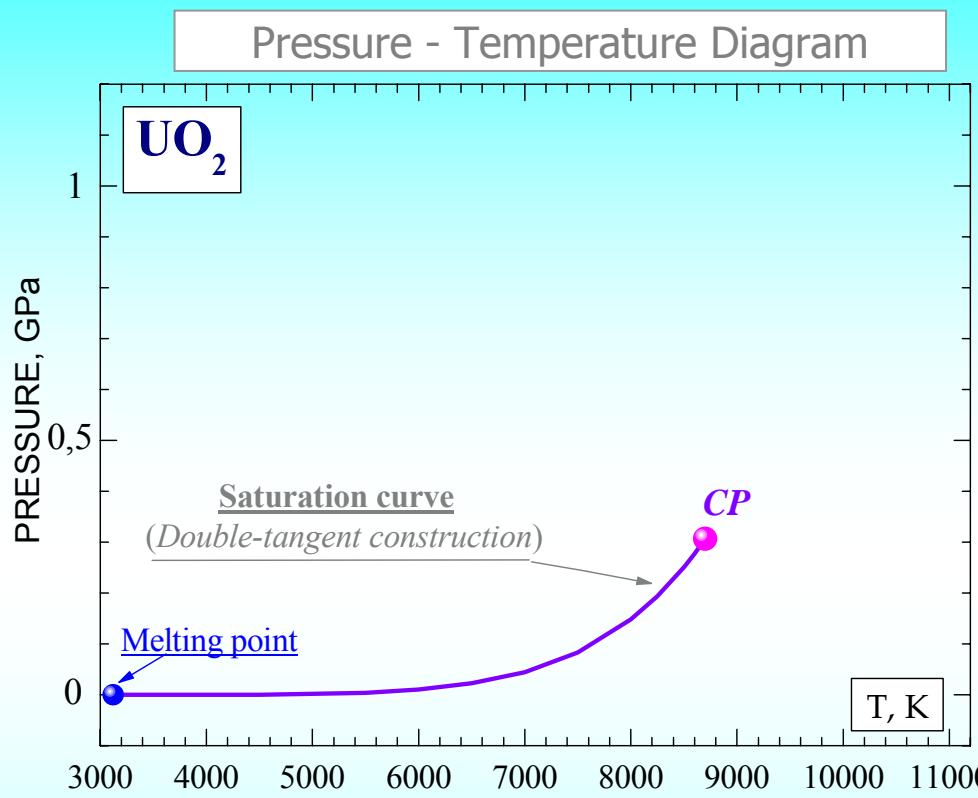
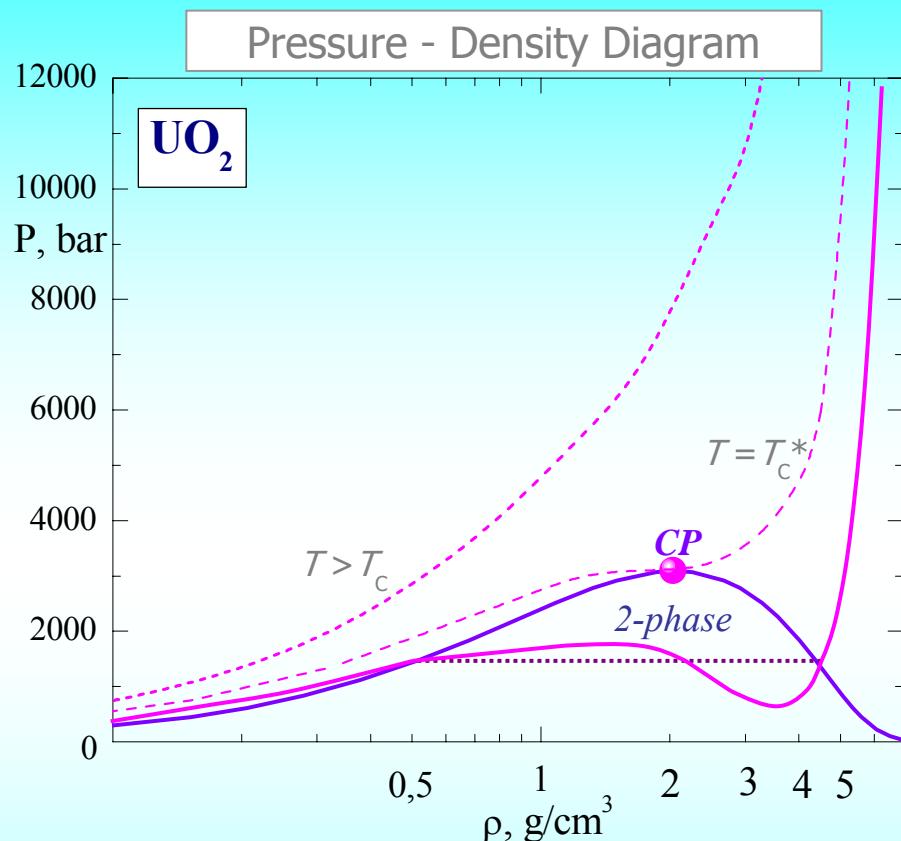
- Hard-sphere Mixture with Varying Diameters
- Modified Mean Spherical Approximation (MSAE+DHSE)
- Modified Thermodynamic Perturbation Theory {TPT- $\sigma(T)$; $\varepsilon(T)$ }

* Iosilevskiy I., Yakub E., Hyland G., Ronchi C. *Int. Journal of Thermophysics* **22** 1253 (2001),

* Iosilevskiy I., Gryaznov V., Yakub E., Ronchi C., Fortov V. *Contrib. Plasma Phys.* **43**, N 5-6 316 (2003)

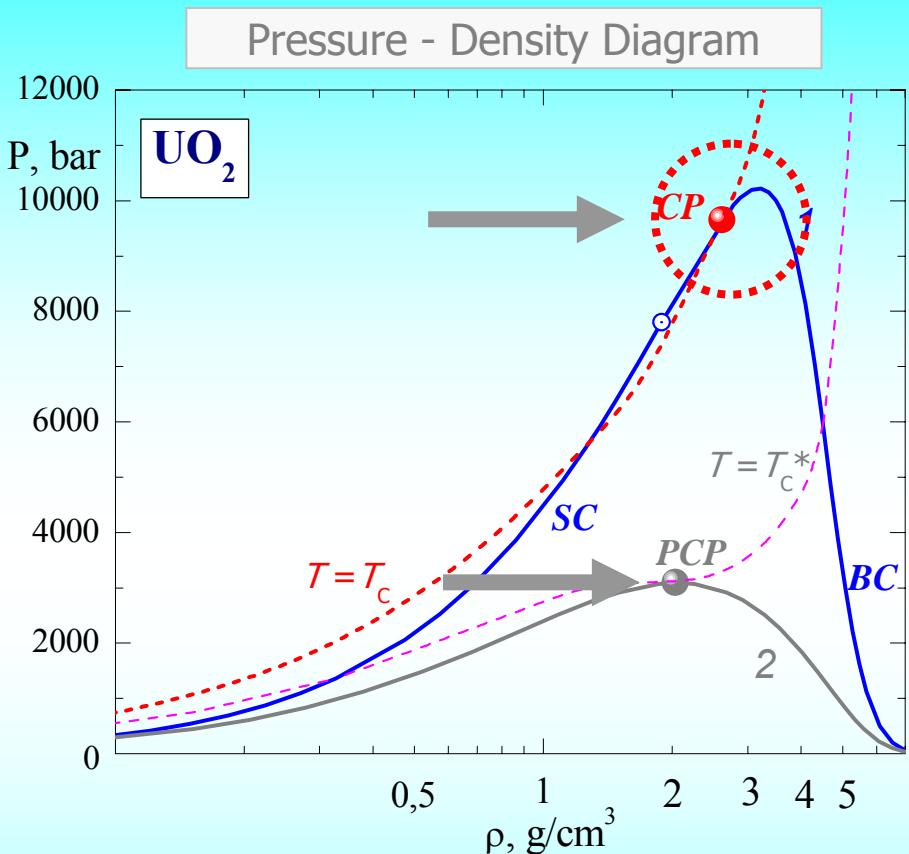
* Ronchi C., Iosilevskiy I., Yakub E., *Equation of State of Uranium Dioxide / Springer, Berlin, (2004)*

Forced-congruent evaporation in U-O system

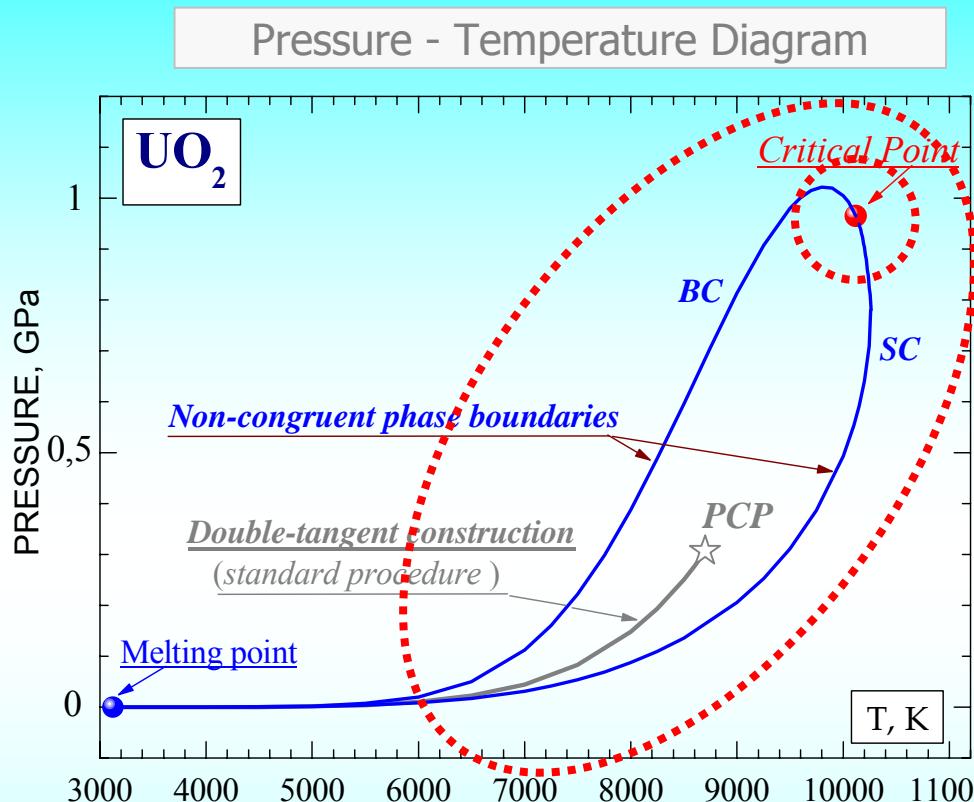


- Stoichiometry of coexisting phases are equal: $x' = x''$
- Van der Waals loops (at $T < T_c$) corrected via the “double tangent construction”
- Standard phase equilibrium conditions:
 $P' = P'' \quad // \quad T' = T'' \quad // \quad G'(P, T, x) = G''(P, T, x)$ $\mu_i'(P, T, x') = \mu_i''(P, T, x'')$
- Standard critical point:
 $(\partial P / \partial V)_T = 0 \quad // \quad (\partial^2 P / \partial V^2)_T = 0 \quad // \quad (\partial^3 P / \partial V^3)_T < 0$

Non-congruent evaporation in U-O system



- 1 – Non-congruent (total) equilibrium
2 – Forced-congruent (partial) equilibrium



- BC – Boiling (liquid) conditions
SC – Saturated (vapor) conditions

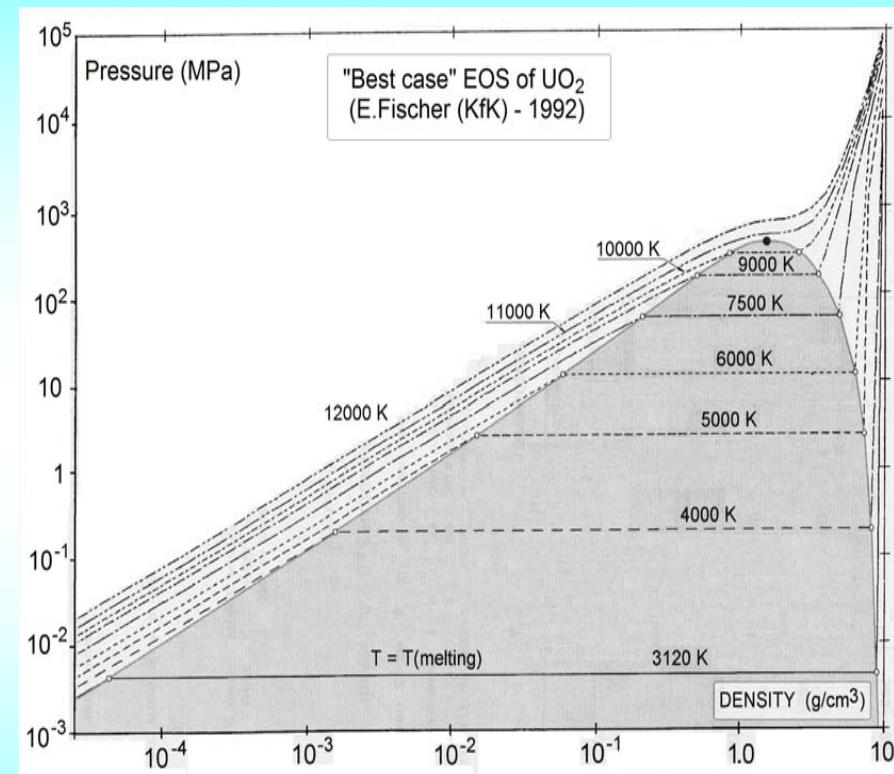
NB! 2-dimensional two-phase region instead of standard P - T saturation curve

NB! High pressure level of non-congruent phase decomposition

NB! Critical point should be of non-standard type: $(\partial P / \partial V)_T \neq 0$ $(\partial^2 P / \partial V^2)_T \neq 0$
It should be instead: $(O/U)_{\text{liquid}} = (O/U)_{\text{vapor}}$ $\{\partial \mu_i / \partial n_k\}_{T,p} = 0$

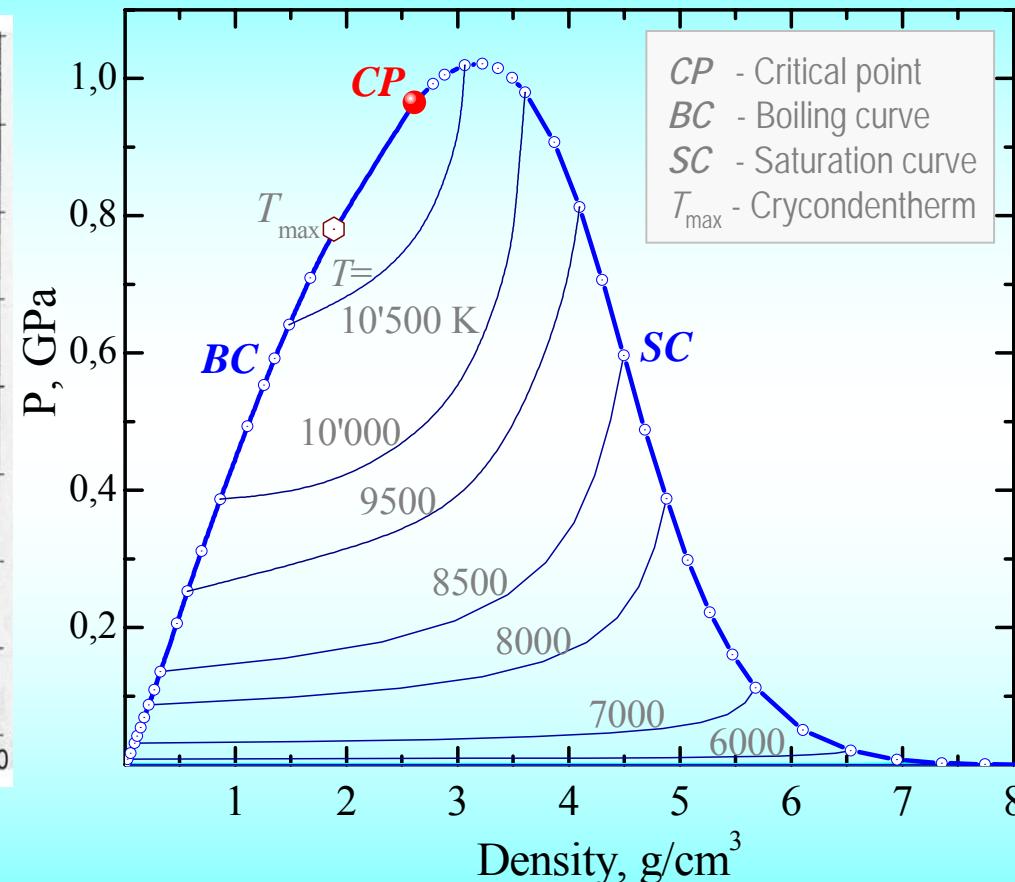
Isotherms in Two-Phase Region

Standard pressure-density diagram



Fischer E.A. Journal of Nuclear Science
and Engineering. 101 97 (1989)

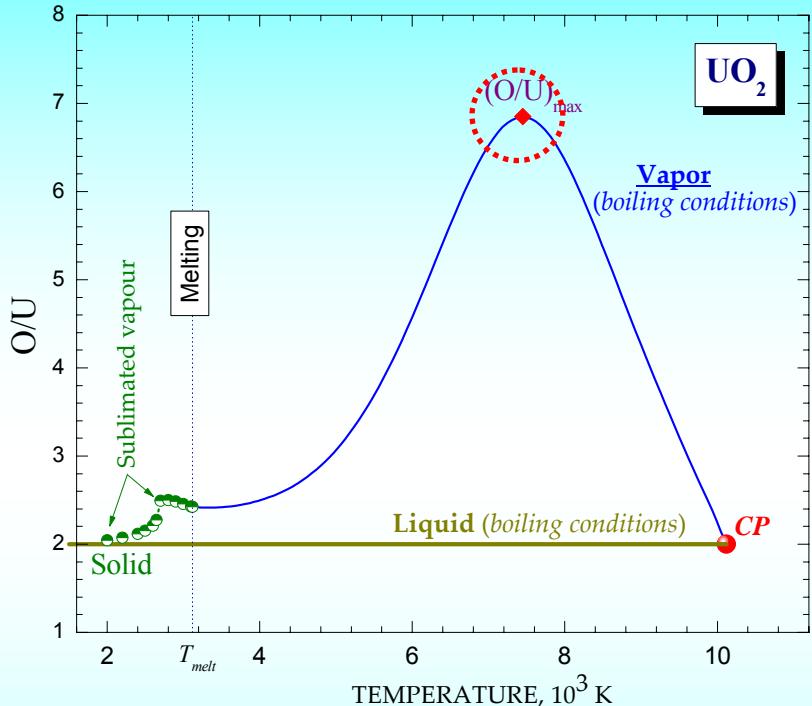
Non-congruent phase transition



- Isothermal phase transition starts and finishes at *different pressures*
- Isobaric phase transition starts and finishes at *different temperatures*

Non-congruent evaporation in U – O system

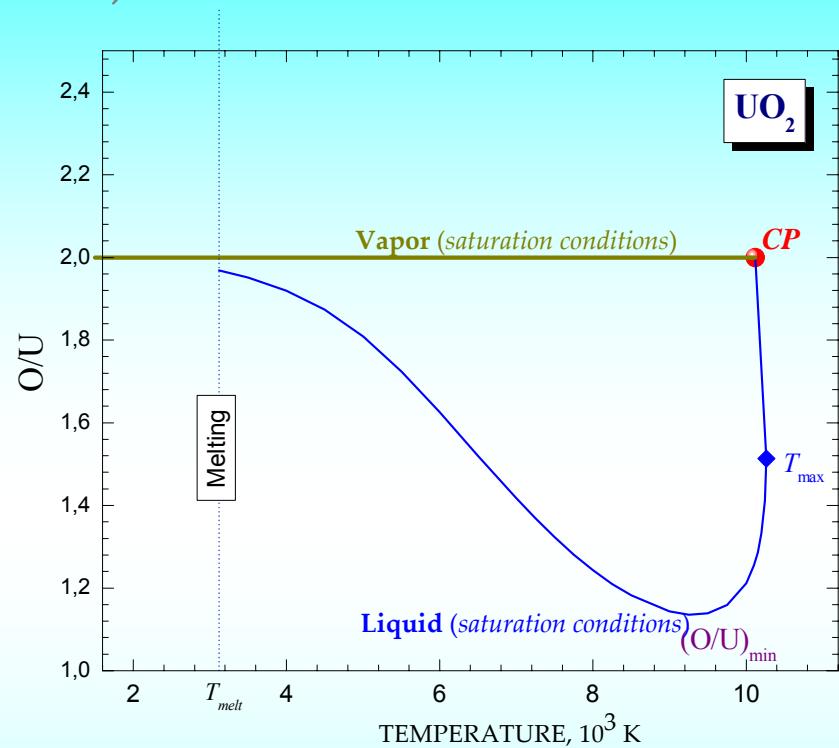
Stoichiometry of Coexisting Phases (two limits)



Boiling Conditions

Liquid (O/U = 2.0) \Leftrightarrow Vapor (O/U > 2.0)

First vapor bubbles over the boiling UO_{2.0}
(oxygen enriched)



Saturation Conditions

Vapor (O/U = 2.0) \Leftrightarrow Liquid (O/U < 2.0)

First liquid drops in vapor UO_{2.0}
("dew point")

NB! High oxygen enrichment of vapor over the boiling UO(2.0)

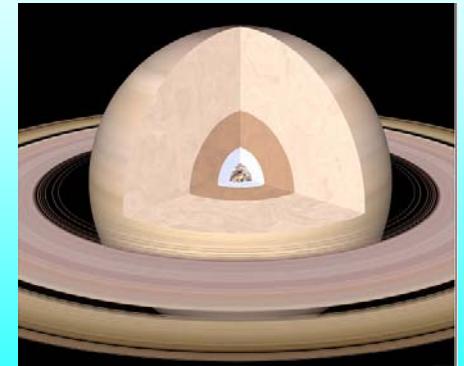
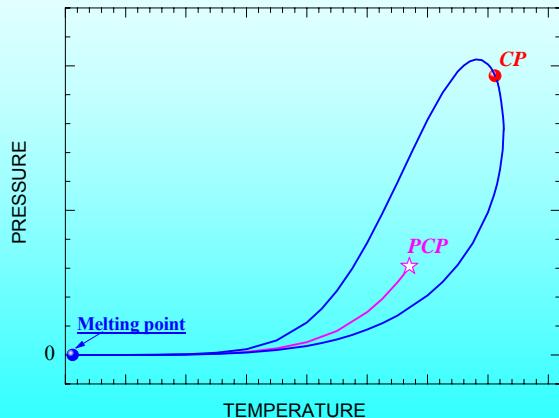
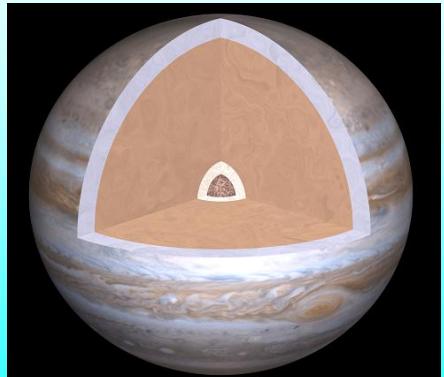
Non-congruence of phase transition in U-O system – – is it an exclusion or a general rule ?

Basic conclusion

- Any phase transition in a system of **two or more chemical elements** must be non-congruent.
- Congruent phase transitions are exclusion.

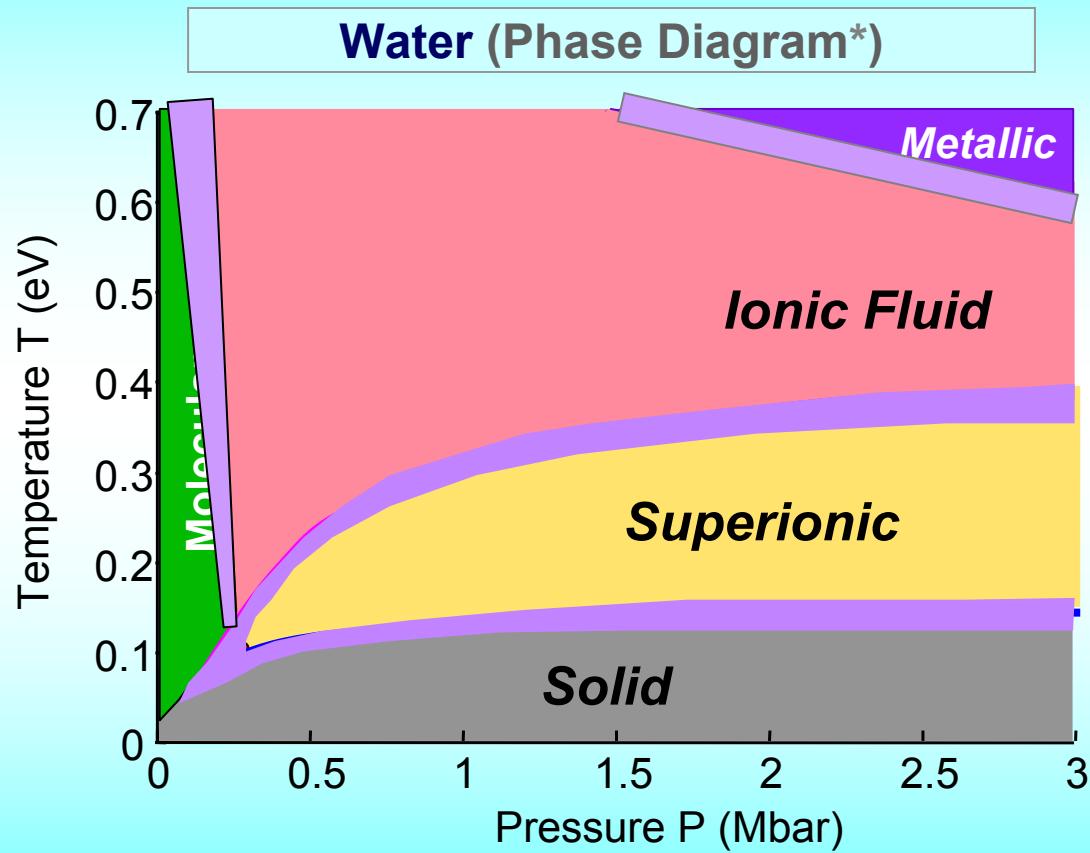
- Hypothetical example of non-congruent phase transition

- “*Plasma Phase Transition*” (PPT) in H_2/He mixture in *Jupiter*, *Saturn* (GP), *brown dwarfs* (BD) and *extra-solar giant planets* (EGP).

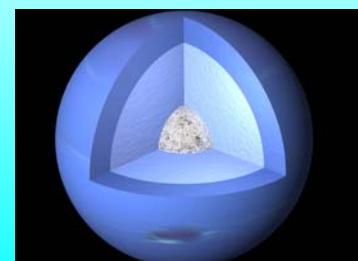
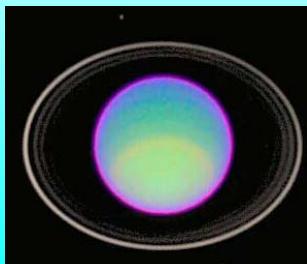


Non-congruent phase transitions in astrophysical objects

Neptune

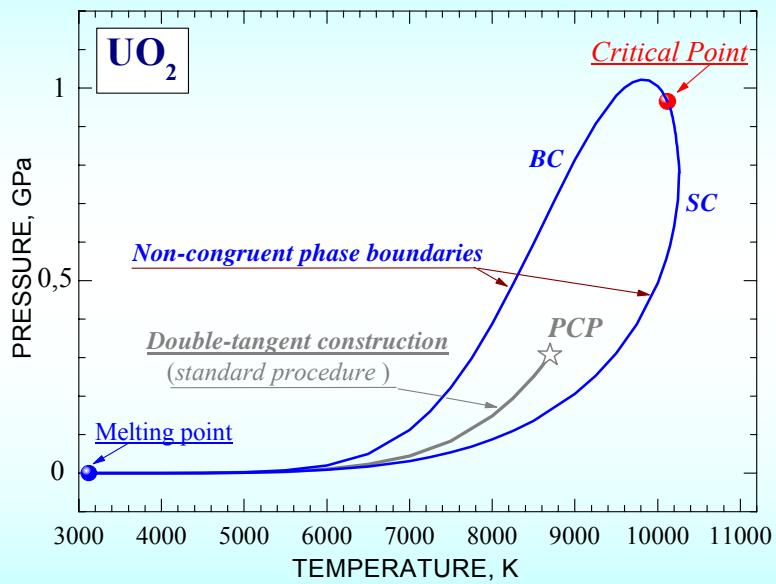


(*) C. Cavazzoni *et al.*, 1999



Hypothetical non-congruent plasma phase transition in ($\text{H}_2 + \text{He}$) mixture in interiors of Jupiter and Saturn

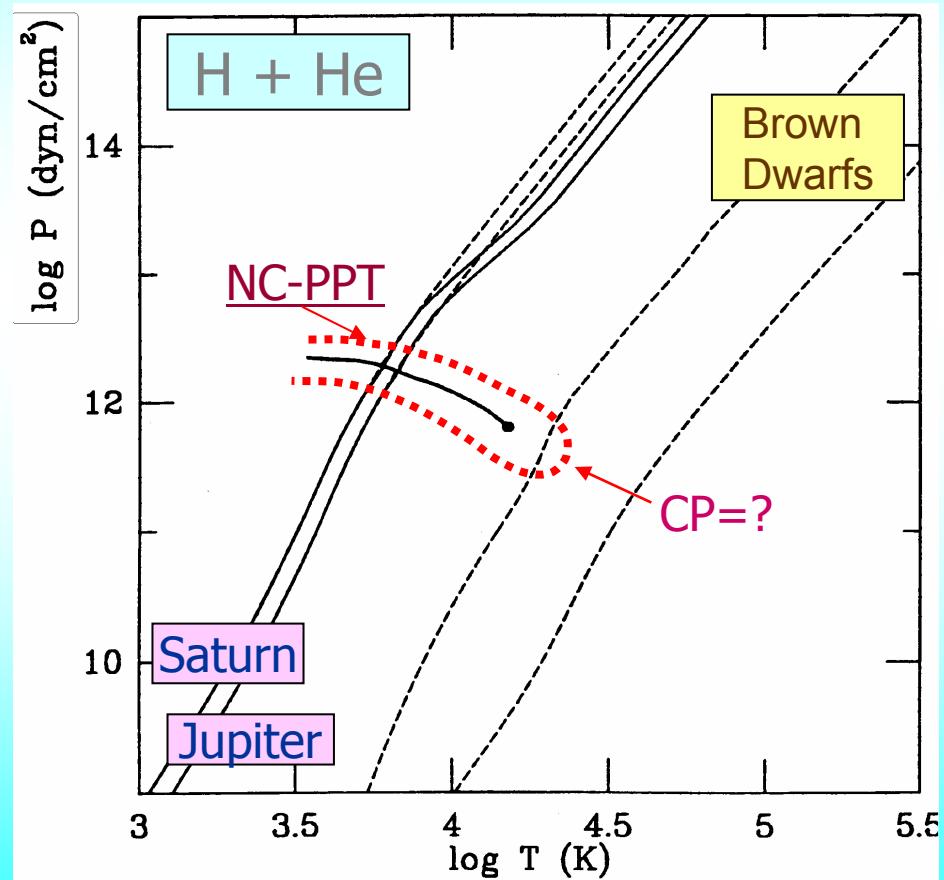
Non-congruent phase transition in U-O system



NB!

Two-phase region in H₂/He must be non-conventional **two-dimensional** domain instead of one-dimensional curve

Non-congruent PPT in $\text{H}_2\text{-He}$ system (NC-PPT)

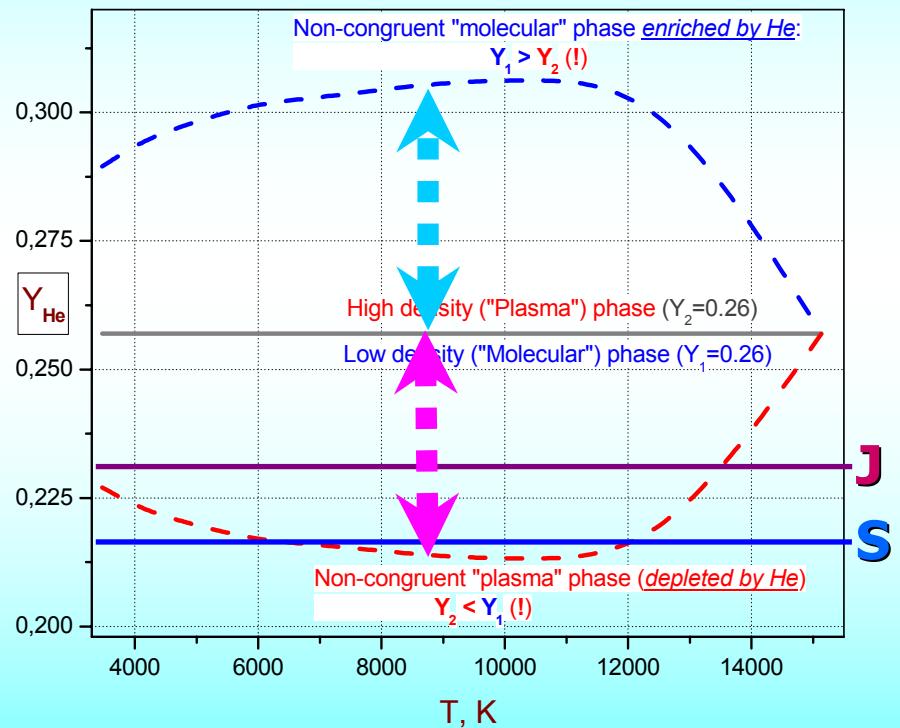


Estimated non-congruence for plasma phase transition in H₂/He mixture of Jupiter and Saturn

(PPT-variant of Saumon, Chabrier and Van Horn – 1995)

Вопрос:

- Является ли оцененная величина гелиевого обогащения (обеднения) пренебрежимо малой, или же заметной?



Phase Separation in Giant Planets:

Jonathan J. FORTNEY, William B. HUBBARD
Icarus, 164 (1) 2003

Atmospheric elemental abundances in Jupiter and Saturn (mass fractions)

Element	SOLAR	JUPITER Galileo	SATURN Voyager	SATURN revised
H	0.736	0.742	0.92	0.76
He	0.249	0.231 ± 0.04	0.06 ± 0.05	0.215 ± 0.035

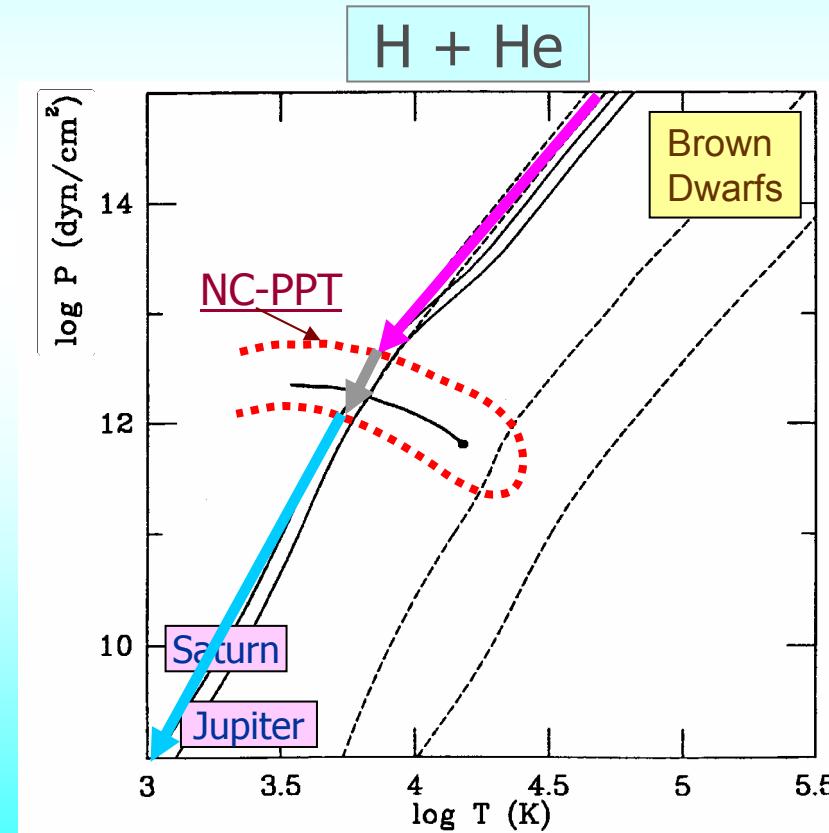
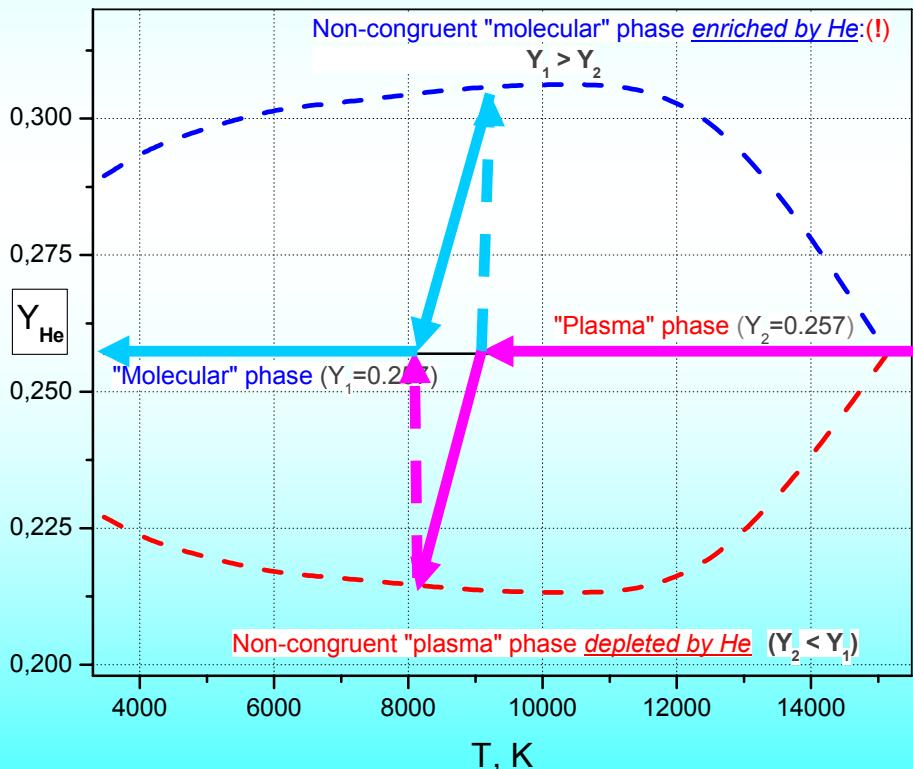
- * Результаты оценки гипотетической неконгруэнтности ПФП в версии Saumon & Chabrier оправдывают **полномерный расчет** этого эффекта.
- Это справедливо для **всех вариантов** гипотетических фазовых переходов, предсказываемых в чистом водороде и гелии, когда эти ФП переносятся в смесь H₂/He.

Estimated non-congruence for plasma phase transition in H₂/He mixture of Jupiter and Saturn

(PPT-variant of Saumon, Chabrier and Van Horn – 1995)

Assumptions:

- Helium is not ionized.
- Atomic helium interacts with neutral hydrogen species only (H₂ and H).
- Interaction of atomic helium with charged species are low and repulsive.



A. Ukrainets & I. Iosilevskiy

in "Physics of matter under extreme conditions",
Ed. V.Fortov, Moscow, IPCP (2005) 116. (in Russ.)

**Assume we know thermodynamics of pure H₂ and He:
How could we obtain the thermodynamics of H₂ + He mixture?**

“Additivity approximation” is widely used for this purpose:

$$\text{EOS}(\text{H}_2 + \text{He}) = x_{\text{H}_2} \text{EOS}(\text{H}_2) + (1 - x_{\text{H}_2}) \text{EOS}(\text{He})$$

(A-1): Additivity of specific enthalpies ($h = H/M$)

$$h_{(\text{A} + \text{B})}(P, T) = x_A h_{(\text{A})}(P, T) + (1 - x_A) h_{(\text{B})}(P, T)$$

(A-1): Additivity of specific volumes ($v = 1/\rho$)

$$V_{(\text{A} + \text{B})}(P, T) = x_A V_{(\text{A})}(P, T) + (1 - x_A) V_{(\text{B})}(P, T)$$

Main issue for the phase transition problem

$$(\partial V_{(\text{A})} / \partial P)_T = \infty$$



$$(\partial V_{(\text{A}+\text{B})} / \partial P)_T = \infty$$

Critical point(s) and (P, T)-coexistence curve(s) of PT(s)

in H₂/He mixture **are the same** identically

as those of phase transition(s) in pure H₂ and He

Conclusion:

P-T phase diagram of H₂/He mixture in frames of “**additivity approximation**” is **superposition** of P-T phase diagrams of pure hydrogen and helium.



Hypothetical phase transitions in interiors of GP-s and BD-s via “additivity approximation”



Phase diagram of H_2/He mixture in frames of ‘additivity approximation’ is **superposition** of $P-T$ phase diagrams for pure hydrogen and helium.

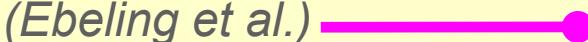
Dissociative Phase Transition in H_2
(*Scandolo S., Bonev S., Militzer B., Galli G.*)



Plasma Phase Transition in H
(*Ebeling et al.*)



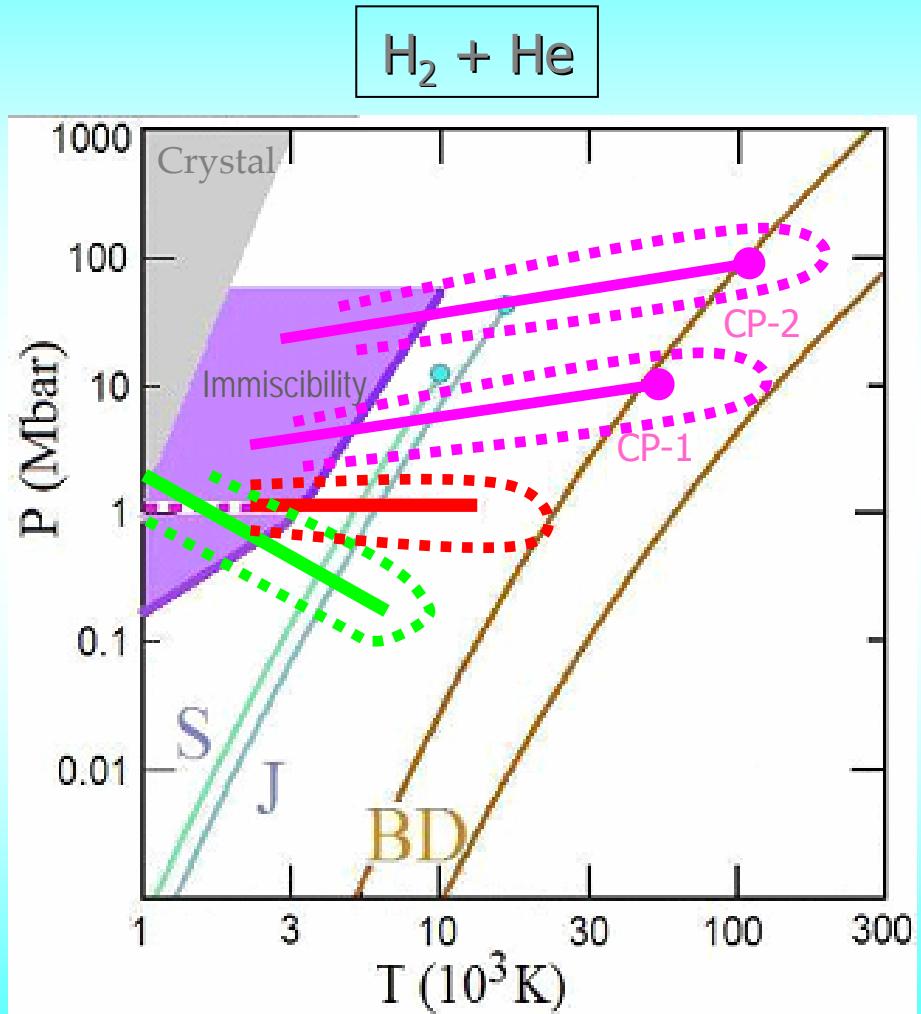
1st Plasma Phase Transition in He
(*Ebeling et al.*)



2nd Plasma Phase Transition in He
(*Ebeling et al.*)



(optimistic)



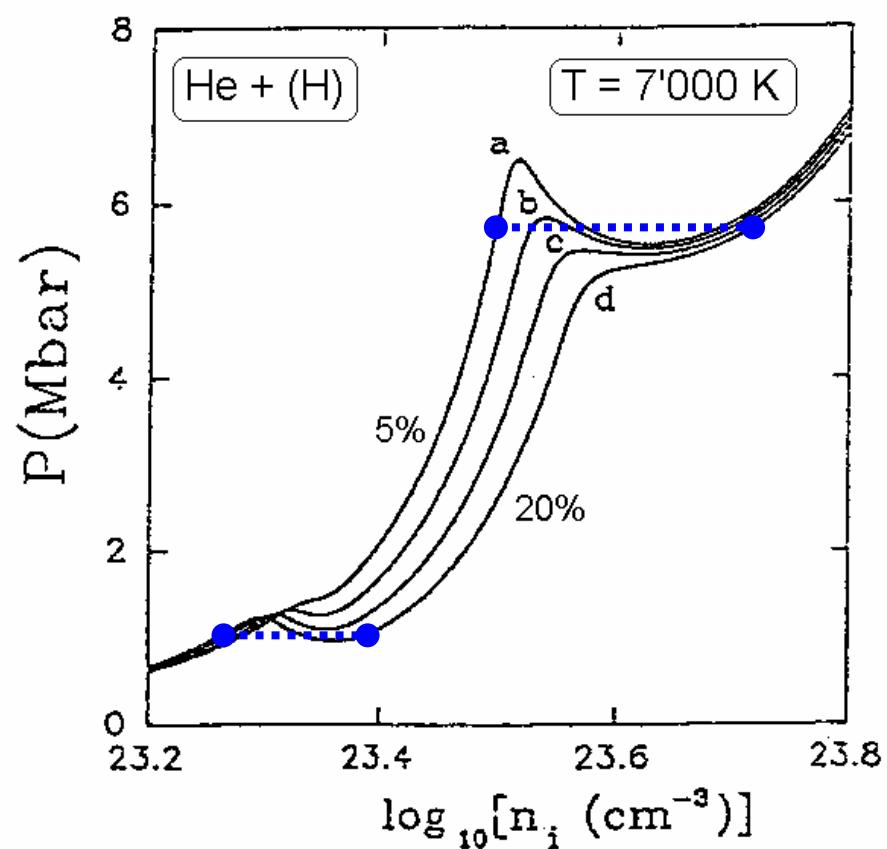
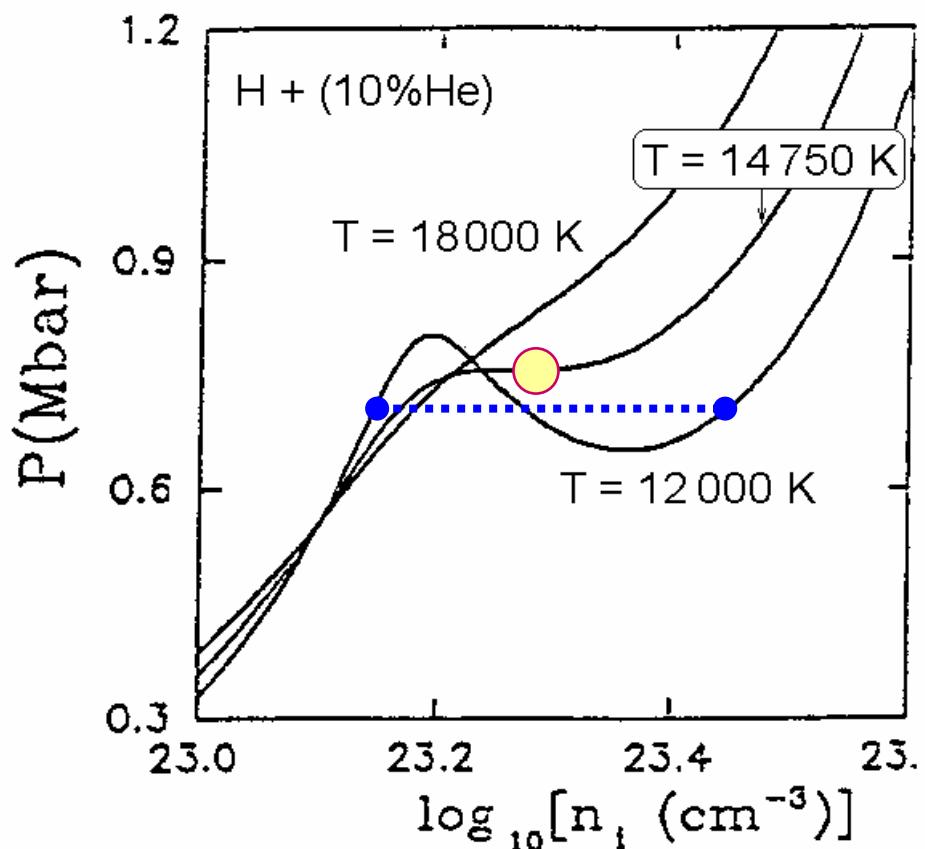
Presence of helium relax phase transition in hydrogen \leftrightarrow presence of hydrogen relax phase transition in helium

Thermodynamics of $\text{H}_2 + \text{He}$ plasma

Contrib. Plasma Phys. 35 (1995) 2, 109–125

Plasma Phase Transition in Fluid Hydrogen-Helium Mixtures

M. SCHLANGE (a), M. BONITZ (b), and A. TSCHITSCHIAN (b)



Thermodynamics of $(H_2 + He)$ plasma (continued)

Contrib. Plasma Phys. 35 (1995) 2, 109 – 125

Plasma Phase Transition in Fluid Hydrogen-Helium Mixtures

M. SCHLANGES (a), M. BONITZ (b), and A. TSCHITSCHJAN (b)

M. SCHLANGES, M. BONITZ , and A. TSCHITSCHJAN

123

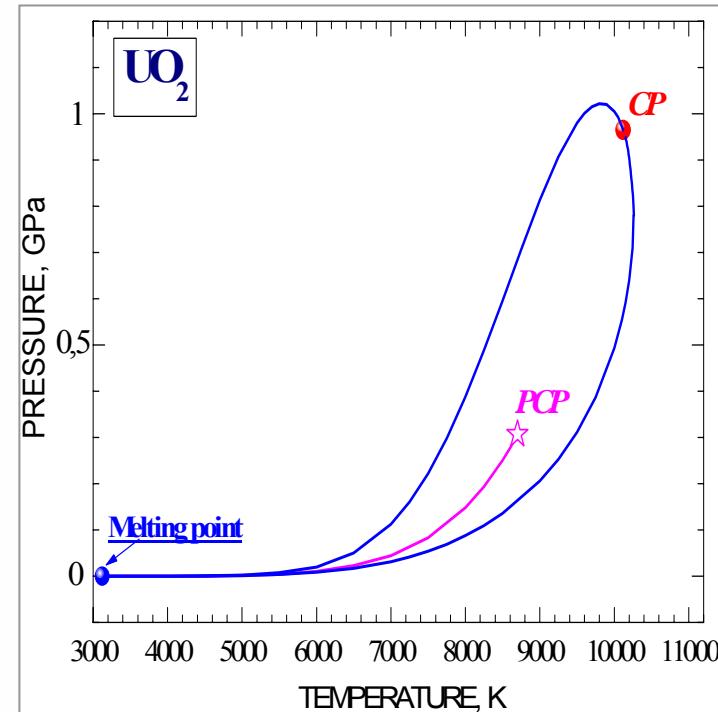
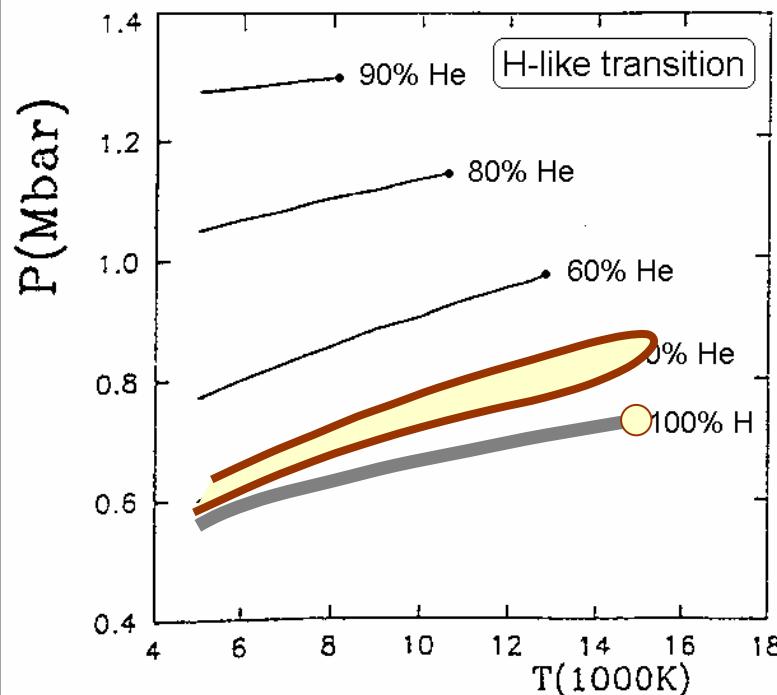


Fig. 7. Coexistence pressure for $H - He$ mixtures for different values of the mixing parameter, for the hydrogen-like plasma phase transition and for the helium-like plasma phase transition.

Elemental Abundance in Solar and Extrasolar Planets

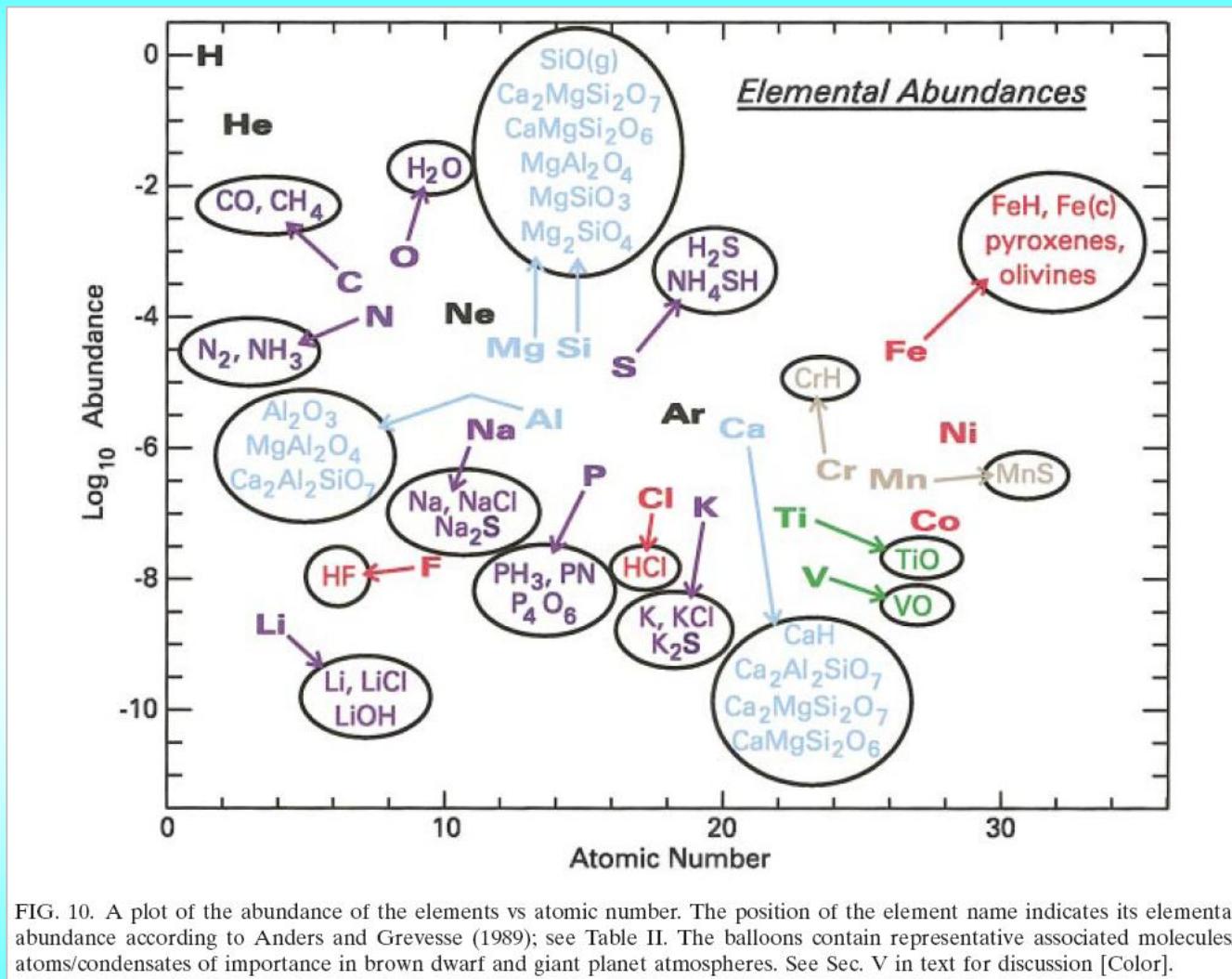


FIG. 10. A plot of the abundance of the elements vs atomic number. The position of the element name indicates its elemental abundance according to Anders and Grevesse (1989); see Table II. The balloons contain representative associated molecules/atoms/condensates of importance in brown dwarf and giant planet atmospheres. See Sec. V in text for discussion [Color].

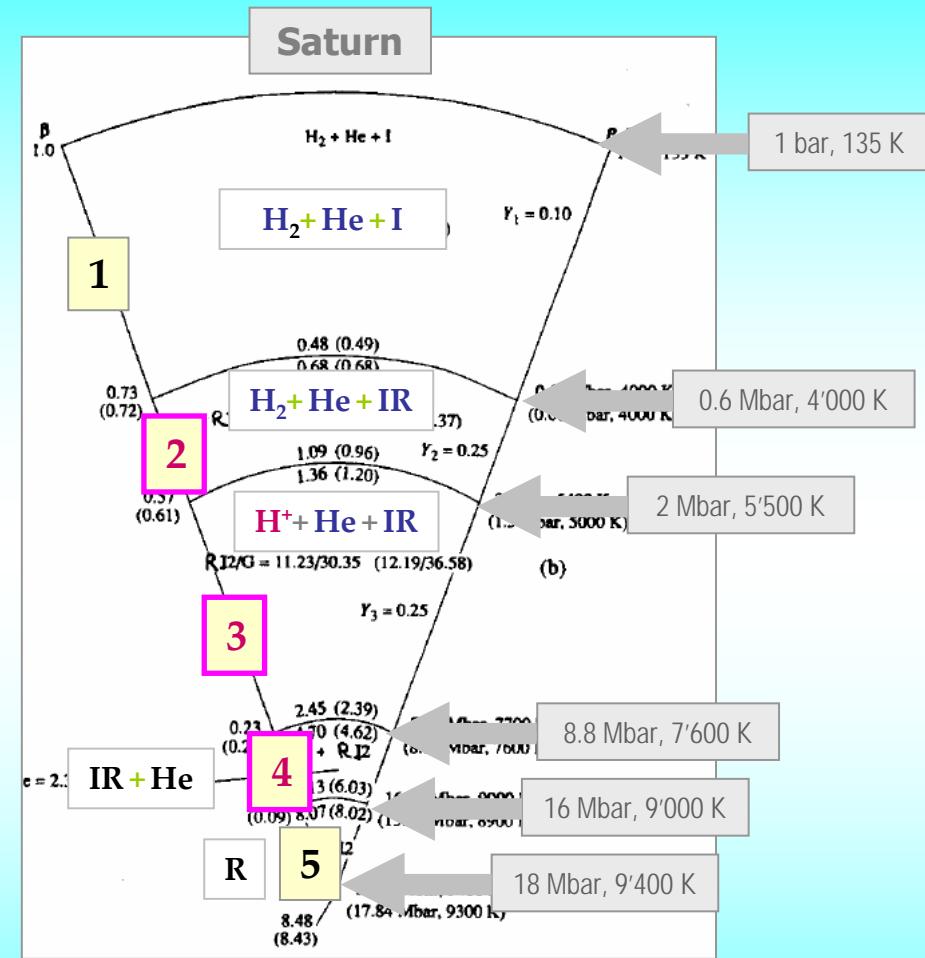
Parameters of the Models of Saturn

Table 4

Parameters of the models of Saturn

Model	Y_0	Y_3	Z_{2-4}	P_m	P_{1-2}	M_{He} , core	M_{core}
$Y_1 = 0.06, Y_2 = 0.25, Z_1 = 0.02, I/R = 2.2$							
MS1	0.267	0.00	0.30	3.0	0.42	10.66	16.18
MS2	0.171	0.00	0.40	3.0	0.64	4.58	8.59
MS3	0.225	0.00	0.30	2.0	0.44	9.88	15.06
MS4	0.133	0.00	0.40	2.0	0.67	4.02	7.65
MS5	0.274	0.25	0.30	3.0	0.46	6.33	9.99
MS6	0.187	0.25	0.40	3.0	0.72	0.05	1.03
MS7	0.285	0.25	0.25	2.0	0.43	7.34	10.74
MS8	0.244	0.25	0.30	2.0	0.55	4.05	6.74
MS9	0.322	0.35	0.25	3.0	0.36	7.66	11.16
MS10	0.278	0.35	0.30	3.0	0.49	3.80	6.38
MS11	0.237	0.35	0.35	3.0	0.62	0.41	1.58
MS12	0.293	0.35	0.25	2.0	0.48	3.90	6.15
MS13	0.255	0.35	0.30	2.0	0.62	0.81	2.11
MS14	0.282	0.35	0.25	1.5	0.60	2.60	4.42
MS15	0.249	0.35	0.30	1.5	0.75	0.007	0.76
$Y_1 = 0.10, Y_2 = 0.25, Z_1 = 0.02, I/R = 2.2$							
MS16	0.275	0.00	0.30	3.0	0.48	10.91	16.54
MS17	0.186	0.00	0.40	3.0	0.73	4.98	9.25
MS18	0.234	0.00	0.30	2.0	0.52	10.22	15.55
MS19	0.149	0.00	0.40	2.0	0.77	4.50	8.45
MS20	0.282	0.25	0.30	3.0	0.53	6.69	10.51
MS21	0.202	0.25	0.40	3.0	0.82	0.57	1.9
MS22	0.277	0.025	0.27	2.0	0.56	6.39	9.7
MS23	0.254	0.25	0.30	2.0	0.64	4.49	7.36
MS24	0.277	0.25	0.25	1.5	0.60	6.64	9.8
MS25	0.263	0.25	0.27	1.5	0.66	5.45	8.41
MS26	0.327	0.35	0.25	3.0	0.43	7.93	11.52
MS27	0.287	0.35	0.30	3.0	0.56	4.16	6.89
MS28	0.248	0.35	0.35	3.0	0.71	0.90	2.34
MS29	0.301	0.35	0.25	2.0	0.57	4.39	6.80
MS30	0.266	0.35	0.30	2.0	0.73	1.36	2.89
MS31	0.291	0.35	0.25	1.5	0.71	3.13	5.12
MS32	0.259	0.35	0.30	1.5	0.87	0.48	1.64

T.V. Gudkova, V.N. Zharkov / Planetary and Space Science 47 (1999) 1201–1210



I – “Ices” ($\text{H}_2\text{O}, \text{NH}_3, \text{CH}_4$)

Y – весовая доля гелия

Z_G – весовая доля $\{\text{H}_2\text{O}, \text{NH}_3, \text{CH}_4 + \text{Fe} + \text{Ni}\}$

Optimized models of Jupiter and Saturn

	JUPITER	SATURN
$M (\oplus)$	317.7	95.1
$M_c (\oplus)$	5	1
M_{ice}/M_c	0.50	0.95
P_c (Mbar)	67.4	15.5
T_c (K)	22600	11900
P_{PPT} (Mbar)	1.71	1.93
T_{PPT} (K)	6880	6070
Y_I	0.29	0.25
Y_{II}	0.326	0.73

TABLE 1

Optimized models of Jupiter and Saturn

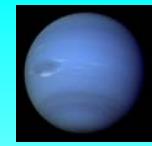
GIANT PLANETS AND THE PLASMA PHASE TRANSITION OF HYDROGEN

D. Saumon, G. Chabrier, W. B. Hubbard, and J. I. Lunine

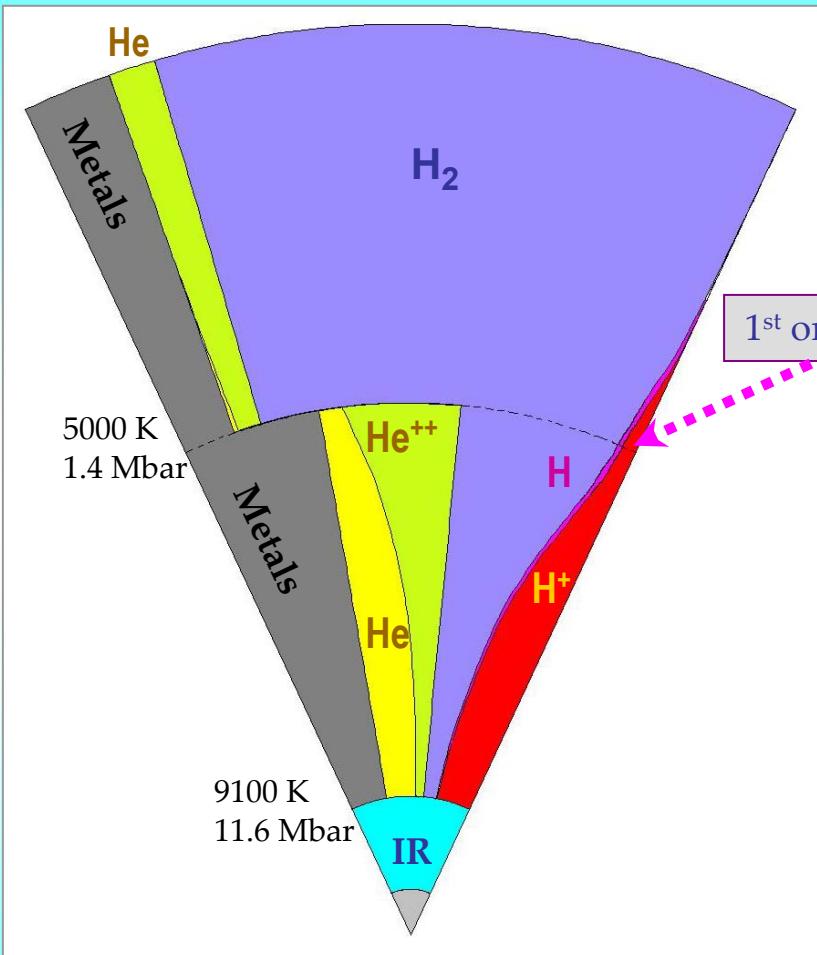
111



Giant planets interior composition



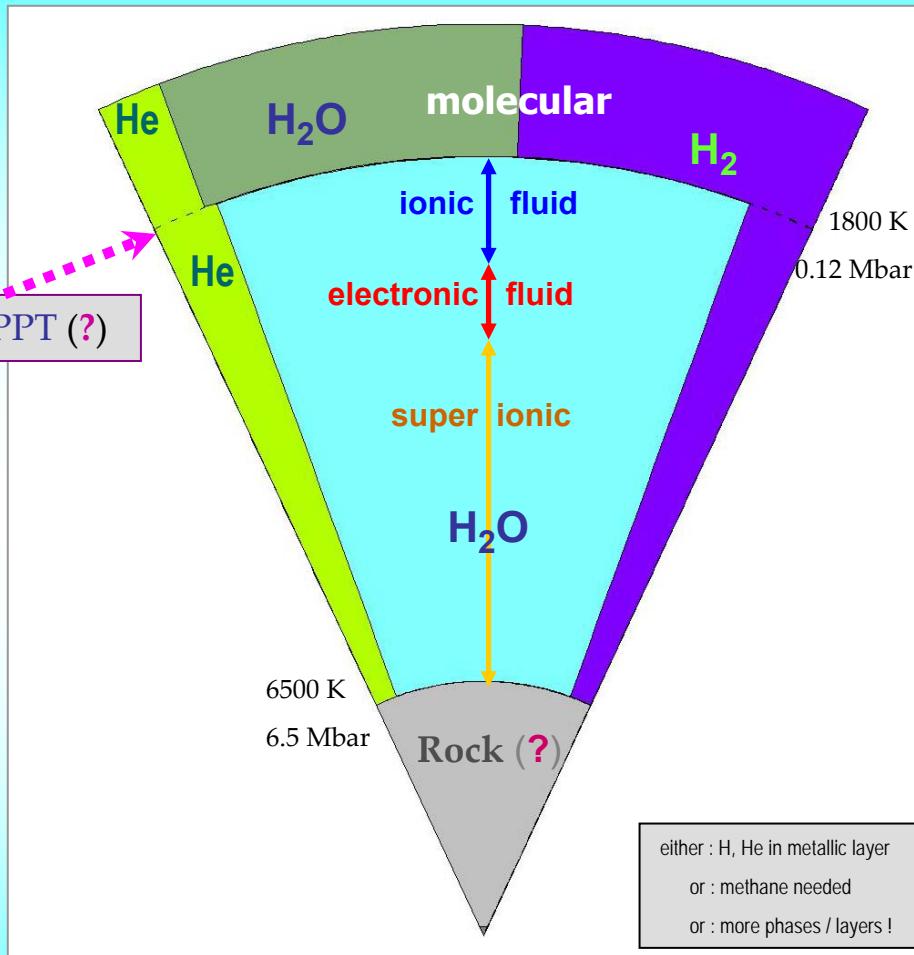
(After N. Nettelmann, R. Redmer, et al., PNP-12, Darmstadt, 2006)



Saturn interior composition
using SCVH-95_EOS

(H₂ + He) Saumon, Chabrier & VanHorn (1995)

Ice-Rock core: Hubbard & Marley, *Icarus* 78, (1989)



Neptune interior composition
using Sesame-EOS (G.Kerley, LANL)

(H₂, He, H₂O) EOS from Sesame-EOS-Tables

H₂O phases from DFT-QMD: T. Mattsson

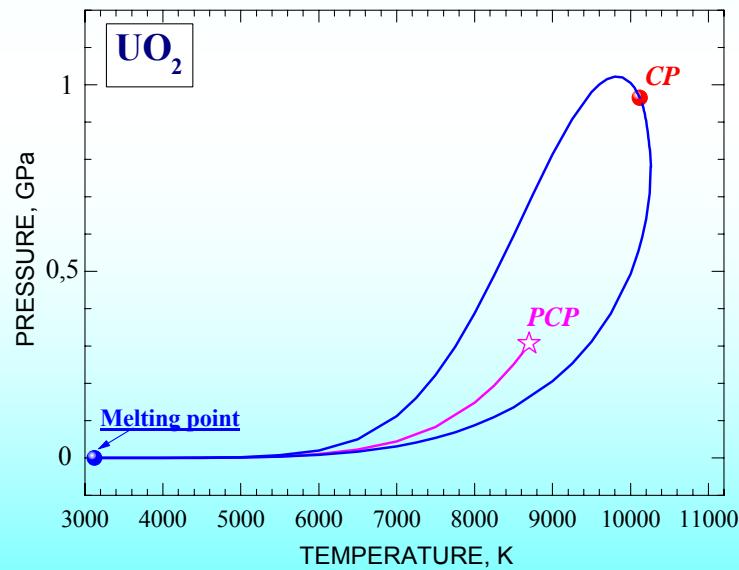
Conclusions and perspectives

- Non-congruence of phase transitions in H₂/He mixture can ‘provoke’ to the H \leftrightarrow He separation in interiors of jovian and extrasolar planets and brown dwarfs.
- First estimation of non-congruence for SCVH-variant of plasma phase transition in H₂/He mixture approves considering of non-congruence in study of helium sedimentation in interiors of Jupiter and Saturn.
- *Ab initio* approaches are very promising for direct numerical simulation of discussed non-congruence for phase transitions in H₂/He mixture in conditions of jovian and extrasolar planets and brown dwarfs.
- New experiments are desirable for study of discussed non-congruence for phase transition in H₂/He mixture under conditions of jovian and extrasolar planets and brown dwarfs.

Clearly there will be enough challenges
to keep us all happily occupied for years to come.

Hugh Van Horn (1990)
(*Phase Transitions in Dense Astrophysical Plasmas*)

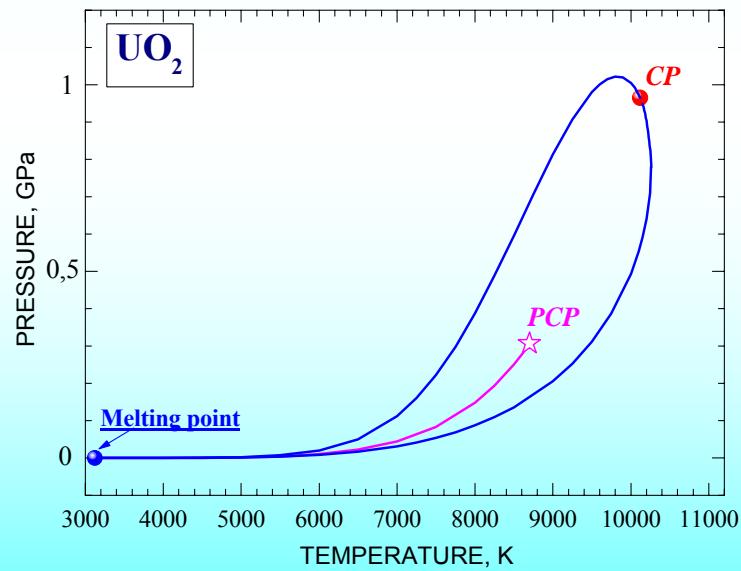
Thank you!



Support: INTAS 93-66 // ISTC 2107 // CRDF № MO-011-0,
and by RAS Scientific Programs

"Physics and Chemistry of Extreme States of Matter" and "Physics of Compressed Matter and Interiors of Planets"

Thank you!



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